We are IntechOpen, the world's leading publisher of Open Access books Built by scientists, for scientists



186,000

200M



Our authors are among the

TOP 1% most cited scientists





WEB OF SCIENCE

Selection of our books indexed in the Book Citation Index in Web of Science™ Core Collection (BKCI)

# Interested in publishing with us? Contact book.department@intechopen.com

Numbers displayed above are based on latest data collected. For more information visit www.intechopen.com



## Thermoplastic Starch (TPS)-Cellulosic Fibers

**Composites: Mechanical Properties and Water Vapor** 

## **Barrier: A Review**

Emilio Pérez-Pacheco, Jorge Carlos Canto-Pinto, Víctor Manuel Moo-Huchin, Iván Alfredo Estrada-Mota, Raciel Javier Estrada-León and Luis Chel-Guerrero

Additional information is available at the end of the chapter

http://dx.doi.org/10.5772/65397

#### Abstract

Current research studies have been focusing on the procurement of environmentally friendly materials, with the aim of resolving the problems created by materials derived from petroleum. Starch is a promising biopolymer for producing biocomposite materials because it is renewable, completely biodegradable, and easily available at a low cost. Thermoplastic starch (TPS), by itself, exhibits poor mechanical properties such as low tensile strength and severe deformations, which limits its application in packaging or films. In addition, TPS presents high hygroscopicity. The use of reinforcing agents in the starch matrix is an effective means to overcome these drawbacks and several types of biodegradable reinforcements, such as cellulosic fibers, whiskers, and nanofibers, have been utilized to develop new and inexpensive starch biocomposites. This chapter provides the latest advances in green composite materials based on TPS and cellulose fibers and includes information on compositions, preparations, and the properties of "green" composite materials elaborated from TPS and cellulose fibers, with the focus on using undervalued natural resources.

**Keywords:** Thermoplastic Starch, Cellulose Fibers, Sustainable Green, Biopolymer, Natural Resource



© 2016 The Author(s). Licensee InTech. This chapter is distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/3.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited. CC BY

## 1. Introduction

Current research studies have been focusing on the procurement of environmentally friendly materials with the aim of resolving the problems created by materials derived from petroleum. Starch is a promising biopolymer used in the production of biocomposite materials because it is renewable, completely biodegradable, and easily available at a low cost. Starch, in the form of its thermoplastic derivate (TPS), has been revealed as an appropriate candidate to be employed as a substitute of synthetic polymers traditionally used for packaging. Starch is not a real thermoplastic polymer, but can be processed after its gelatinization by mixing it with enough water and/or plasticizers [1, 2]. In most investigations the plasticization of the material is carried out by casting a dispersion of starch with glycerol [1, 3].

TPS films are reported to have low permeability to gases, poor water vapor barrier properties, and must resist the tensions arising from their use in packaging [4].

Therefore, improving the resistance of TPS films to traction forces is a factor that must be taken into consideration for their use [5, 6]. The use of reinforcing agents in the starch matrix is an effective means of overcoming these drawbacks and several types of biodegradable reinforcements such as cellulosic fibers, whiskers, and cellulose nanofibers (CNF) have been utilized to develop new and inexpensive starch biocomposites. Cellulose is the most abundant, renewable polymer in the world; it is found in plant cell walls and it can also be synthesized by some bacteria. Its reinforcing property is remarkable [7]. Basically two types of nanoreinforcements can be obtained from cellulose: microfibrils (or CNF) and whiskers.

Improving the mechanical and water barrier properties by the addition of CNF (extracted from different botanical sources) can depend on the correct dispersion and the generation of an active nanoreinforcement/matrix interface.

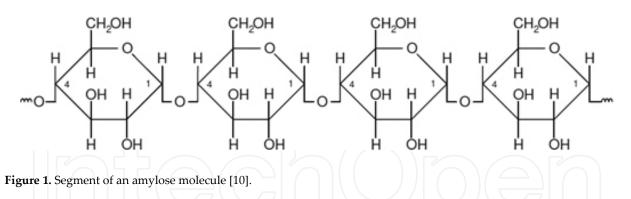
This chapter provides the latest advances in green composite materials based on TPS and cellulose fibers and includes information on compositions, preparations, and the properties of "green" composite materials elaborated from TPS and cellulose fibers, with the focus on using undervalued natural resources.

## 2. Starch

Starch is a biodegradable and widely available natural resource [8], and constitutes the main source of carbohydrate reserves in plants. This polysaccharide is found in different parts of the plants and can be isolated from seeds, fruits, leaves, tubers, and roots [9].

The functionality of starch is largely due to its two components of high molecular weight: amylose and amylopectin. Most of the starches contain between 20 and 30% of amylose and 70 and 80% of amylopectin; these proportions depend on the plant source. Amylose molecules consist of approximately 200–20,000 glucose units joined by  $\alpha$ -1,4 glycoside bonds (**Figure 1**) in unbranched chains or coiled helix [10].

Thermoplastic Starch (TPS)-Cellulosic Fibers Composites: Mechanical Properties and Water Vapor Barrier: A Review 87 http://dx.doi.org/10.5772/65397



The structure of amylopectin is different from that of amylose; amylopectin molecules contain  $\alpha$ -1,4 and  $\alpha$ -1,6 glycosidic bonds, as can be seen in **Figure 2**. The glycosidic bonds join the glucose molecules in the main amylopectin chain. Branches of the main chain are often found, which are due to the  $\alpha$ -1,6 glycosidic bonds with other glucose molecules.

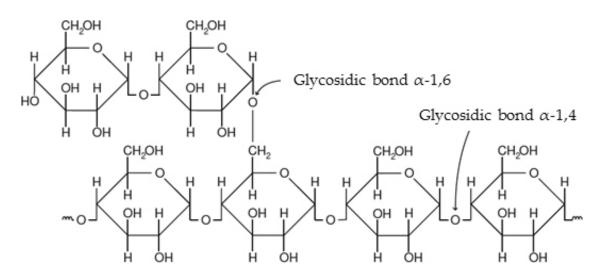


Figure 2. Segment of an amylopectin molecule [10].

The bonding points of the branches constitute between 4 and 5% of all the bonds [11]. Amylopectin molecules are significantly larger than the amylase molecules; some contain between 10,000 and 20,000,000 glucose units. The molecular weight of amylose is between 0.1 and 1 million g/mol while that of amylopectin ranges between 10,000 and 1000 million g/mol [10].

One of the most important properties of natural starch is its semicrystallinity and amylopectin is the dominant component for crystallization in most starches. The significant commercial properties of starch, such as mechanical resistance and flexibility, depend on the resistance and character of the crystalline region, which in turn depend on the amylose/amylopectin ratio and thus on the type of plant; these properties also depend on the distribution of molecular weight, degree of branching, and the conformation process of each polymeric component [12].

Starch is versatile due to its variety of uses; it is also one of the most important ingredients at an industrial level. In the food industry, starch is used to provide a wide range of functional

properties and is probably the most utilized hydrocolloid [13]. It can be used in flavor encapsulation, as a thickening agent or a filling agent, in bakery products, production of syrups, etc. Starch is also included in many other industries such as textiles, paper, cosmetics, plastics, pharmaceutical, and adhesives.

Currently, the negative environmental impact caused by synthetic polymer wastes, denominated plastic materials, is well known and there is now a growing interest in biodegradable materials like starch to substitute the conventional plastic materials, such as polyethylene and polystyrene. A number of studies have reported the use of starch in the manufacture of fast food utensils and packaging material [14].

It is clear that one of the alternatives is the use of starch from nonconventional sources, particularly in countries where there is a high production of raw material for the production of this polymer. Nonconventional sources of starch have attracted much attention, given their diversity of properties, which allow their application in different industries, including the food industry.

**Table 1** shows the amylose content, granule size, and gelatinization temperature of a few starches from nonconventional sources such as chestnut [15], kudzu [16], ramon [17], chayote [18], -parota [19], makal [20], sorghum [21], mango [22], and okenia [23], which could be considered for the production of biodegradable materials. In general, the new botanical sources are always widely available in the countries, which produce them.

Type of starch	Amylose (%)	Granule size (µm)	Gelatinization temperature (°C)	Reference
Chestnut	26.6	4–21	61.9	[15]
Kudzu	22	2–20	64-83	[16]
Ramon	25.3	6.5–15	83.05	[17]
Chayote	12.9	7–50	67.7	[18]
Parota	17.5–21.3	20–28	76-78	[19]
Makal	23.6	12.4	78.4	[20]
Sorghum	11.2–28.5	Not reported	70–75	[21]
Mango	9–16	7-28	77–80	[22]
Okenia	26	Not reported	71.3	[23]

Table 1. Characteristics of starches from different botanical sources.

## 3. Thermoplastic starch (TPS)

To produce a film based on starch, a high content of water or plasticizer is required (glycerol, sorbitol). These plasticized materials (application of mechanical and thermal energy) are known as thermoplastic starches [8].

The development and production of TPS is considered to be important for the reduction of the total quantity of synthetic plastic wastes in the world [24].

TPS is a material which is obtained through the structural disruption (modification) occurring inside the starch granule when it is processed with a low water content and the action of shear force and temperature in the presence of plasticizers which do not evaporate easily during the processing [25].

To date, it is known that the techniques (extrusion, injection molding, and film casting) for processing starch-based materials are similar to those used for conventional polymers. However, it is important to note that although the processing of starch is complicated, it can be achieved successfully if an appropriate formulation is developed and adequate processing conditions are established [26].

A simple and well-established technique for producing sheets or films by extrusion is the use of a twin-screw extruder with a slit or flat film die, followed by a takeoff device for orientation and collection [27, 28].

Foaming extrusion has mainly been used to produce loose-fill packaging materials, in a similar way to the production of extruded expanded snack foods [29].

Twin-screw extrusion is the most widely used and is preferred because of its ease of feeding, longer residence time, more extensive shear, and more flexible temperature control [30, 31].

The high viscosity and poor flow properties of starch-based materials present difficulties during injection molding, while the lack of reliable parameters makes it difficult to design the optimum processing conditions [26].

Compression molding has been intensively investigated for processing starch-based plastics, particularly in the production of foamed containers, and generally involves starch gelatinization, expanding, and drying. Apart from gelatinization agents, mold-releasing agents such as magnesium stearate and stearic acid are also often used in formulations to prevent the starch sticking to the mold [26].

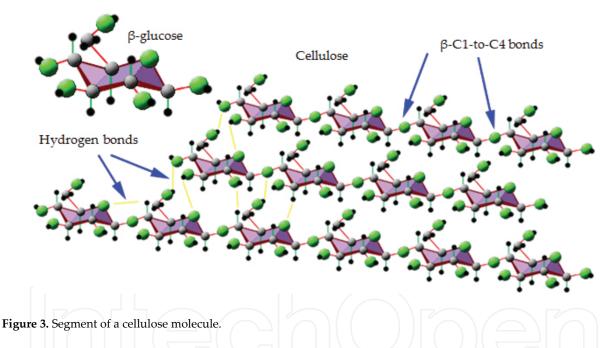
The casting technique for preparing starch films includes the preparation of a dispersion, gelatinization at 95°C, casting in acrylic or Teflon plates, and a drying period of approximately 24 h at 40–75°C [32–34]. Glycerol is the most used plastifier in the preparation of starch films. The resulting starch film can have a thickness between 0.02 and 0.10 mm [26].

Over the last few years, much research has focused on the modification of the starch in the production of a good thermoplastic material [35, 36]. A number of mixtures of TPS with biopolymers based on thermoplastic starch are being commercialized with a certain degree of success, by companies such as Mater Bi<sup>®</sup> (Novamont S.P.A., Italia) in Italy [www.mater-bi.com], Carghill-Down<sup>®</sup> in the U.S., and by others in Spain, Germany, France, Japan, Denmark, and Canada [37].

## 4. Cellulose

Cellulose is the most abundant renewable biomass material in nature and is also a major component of plant cell walls. It has been widely used in the form of wood and plant fibers as an energy source, for building materials, and for clothing. Cellulose can be used as feedstocks for producing biofuels, bio-based chemicals, and high value-added bio-based materials. In the past, much attention has been given to the conversion of cellulose-to-cellulose-based composites due to its properties relating to mechanical strength, biocompatibility, biodegradation, and bioactivity, as well as its potential applications which include biomedical, antibacterial, water pretreatment, and in the field of functional materials for photocatalysis [38].

The basic building block of cellulose is  $\beta$ -D-glucose (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>) (**Figure 3**). To form the "cellobiose" molecule, one water molecule is eliminated for every two glucose molecules. Then, from the condensation of various cellobiose molecules, the cellulose molecule is formed. The cellulose chains are found very close to each other due to their hydrophilic character [39].



The elemental composition of cellulose was discovered in 1842 by the French chemist Anselme Payen [40] and cellulose consists of: 44.0–45.0% carbon (C), 6.0–6.5% hydrogen (H), and 48.5–50.0% oxygen (O).

Table 2 presents a list of the chemical composition of several materials containing cellulose [41].

In plants, cellulose can be found in the form of microfibers in the primary, secondary, and tertiary cell walls, organized either unidirectional or as a woven mesh. Cellulose microfibers are 5 nm thick strands of glucopyranose molecular chains, whose Young's module is 134 GPa, with a density of 1.5 g/cm<sup>3</sup>, and a tensile strength (TS) estimated at approximately 2 GPa. Cellulose microfibers have a diameter of 20–200 Å, while the length can reach several dozen microns. These characteristics are responsible for the interesting mechanical properties of the

Composition (%)					
Source	Cellulose	Hemicellulose	Lignin	Extract	
Hardwood	43-47	25–35	16–24	2–8	
Softwood	40-44	25–29	25–31	1–5	
Cotton	95	2	1	0.4	
Bagasse	40	30	20	10	
Coconut fiber (coir)	32–43	10–20	43–49	4	
Corn cobs	45	35	15	5	
Corn stalks	35	25	35	5	
China grass (ramie)	76	17	1	6	
Flax (unretted)	63	12	3	13	
Flax (retted)	71	21	2	6	
Hemp	70	22	6	2	
Jute	71	14	13	2	
Sisal	73	14	11	2	
Kenaf	36	21	18	2	
Sunn	80	10	6	3	

microfibers, which are extracted from the biomass by means of a chemical treatment followed by a mechanical treatment; the aim being to obtain a homogenous suspension of individual microfibers [42, 43].

 Table 2. Chemical composition of materials containing cellulose [41].

In this process, the disintegration of the cellulose was achieved by the generation of high shear forces. As a consequence of the above, the microfibers detach from the surrounding material and remain united to other microfibers, thereby forming a network of fibers with nanometric dimensions. The cellulose obtained from this procedure is denominated cellulose nanofibers. These CNF are packages of fibers on a nanometric scale. It has been determined that, within the plant, these CNF are capable of supporting the pressure exerted by the water contained in the plant cells and thus the interest in using cellulose in this form for nanocomposites, based on the concept of introducing nanometric loads (fillings) in a polymeric matrix [44].

#### 4.1. Extraction of cellulose nanofillers

Several processes have been used to extract highly purified CNF from cellulosic materials. All these methods lead to different types of magnesium stearate and stearic acid, depending on the cellulose raw material and its pretreatment, and more importantly, depending on the disintegration process itself [45]. CNF extraction can be divided into: chemical hydrolysis and extraction by mechanical force.

#### 4.1.1. Extraction by chemical hydrolysis

A commonly used extraction methodology of CNF is acidic hydrolysis of the amorphous regions surrounding the embedded CNF and cleavage of the bundles, followed by filtration or centrifugation to exclude dissolved noncrystalline elements [46, 47]. The methodology is beneficial in that it can be performed on very small quantities of cellulose, it requires only the simplest laboratory equipment, and the CNF can be obtained without any induced imperfections caused by mechanical processing. The conditions typically involve the use of aqueous solutions of sulfuric acid, stirred at 45–60°C at atmospheric pressure until a homogeneous beige solution is obtained [48, 49]. After acid hydrolysis, the suspension is diluted with water in order to stop the chemical reaction. This suspension is subjected to centrifugation to obtain the cellulose and eliminate excess acids. The resulting cellulose is washed with water using centrifugation and is finally dialyzed using membranes to reach a neutral pH [50].

This procedure results in CNF having anionic groups on the surfaces (leading to electrostatic stabilization of the nanocrystals in suspension) with the ability to form chiral nematic liquid crystalline phases in concentrated solutions [51]. The form of cellulose obtained was denoted microcrystalline cellulose (MCC) by Battista in 1975 [52].

Exaggerated hydrolysis can typically be noted as the solutions turn dark or black in color as the degradation of the CNF occurs. This phenomenon was reported by Roman et al. in 2004 [53], who assigned the crystal degradation to potential induced thermal degradation related to the sulfate groups, introduced as a functional surface on the CNF when sulfuric acid is used for hydrolysis.

#### 4.1.2. Extraction by mechanical force

The mechanical methods to extract CNF from wood pulp and parenchyma cells typically involve a high-pressure homogenizer treatment [42, 54], a microfluidizer [55, 56], a high-pressure refiner, a super-grinder treatment [57], or ultrasonication [58]. The form of cellulose obtained was denoted microfibrillated cellulose (MFC) by Herrick et al. and Turbak et al. in 1983 [42, 43]. These processing methodologies have in common the fact that they rely on the application of high shear forces on cellulose fiber suspensions in order to mechanically liberate the CNF from the original plant cell wall structure. In a high-pressure homogenizer this is achieved by allowing a cellulose suspension to pass under high pressure through a thin slit where it is subjected to high shear forces. The shear forces serve to disintegrate the microfibrils or microfibril bundles in the plant cell wall, resulting in CNF with diameters of about 5–100 nm [59].

High-intensity ultrasonication consists of a combination of chemical pretreatment and highintensity ultrasonication. To obtain the cellulose fiber, first, the lignin is eliminated from the samples by immersion for 1 h at 75°C in a solution of sodium chlorite, previously acidified [60, 61]. After the bleaching, the samples are treated with potassium hydroxide (3%) at boiling point for 2 h. After which, they are subjected once again to a treatment with potassium hydroxide at a different concentration (6%), the aim being to eliminate hemicellulose, residual starch, and pectin. After the application of the chemical treatment, the samples are washed with distilled water and the resulting cellulose fibers are immersed in distilled water. A total of 120 ml of this solution containing purified cellulose fibers is placed in an ultrasound generator of 20–25 kHz in frequency equipped with a cylindrical titanium alloy probe tip of 1.5 cm in diameter. The subsequent ultrasonication is conducted for 30 min to isolate the CNF.

#### 4.2. Green composites of TPS and cellulose fibers

Human beings are known to be highly dependent on synthetic polymers derived from petroleum for the elaboration of diverse packaging and utensils, giving rise to environmental problems [45]. However, reports have been published of the combination of natural fibers, which are completely biodegradable, for the development of green composites. These materials are environmentally friendly and their use can reduce contamination on a global level [45].

Nanocomposites contain materials with a nanometric dimension ranging between 1 and 100 nm. These materials surpass conventional composite materials due to their superior thermal, mechanical, and barrier properties [62, 63]. Biodegradable polymers, in particular, may require improvement in terms of brittleness, low thermal stability, and poor barrier properties [63].

The physical and mechanical properties of the polymeric material reinforced with cellulose are strongly dependent on their structure, relaxation, and morphological processes, as well as a good dispersion of the fiber in order to achieve the minimization of holes. Indeed, a good dispersion of fibers in the polymer matrix has been reported as something very difficult to achieve [64].

The processing of TPS reinforced with cellulose fibers processing is similar to most conventional synthetic thermoplastic processing [65]. Most thermoplastic operations involve heating and forming into desired shapes, and then cooling. Processing techniques used on thermoplastics can also be used in the TPS reinforced with cellulose fibers. These include extrusion, injection molding, internal mixing, compression molding, and others [66].

#### 4.3. Preparation of green composites: mechanical properties and water vapor barrier

From a tensile test, basic mechanical properties of a TPS can be obtained such as tensile strength (the maximum tensile stress a TPS can withstand before it breaks), percentage of elongation at breakage (E) (flexibility), percentage of elongation at yield (EY), and modulus of elasticity (EM) (stiffness) [67, 68].

It has been reported that the mechanical properties of a starch film is affected by the glass transition temperature, degree of crystallinity of the films, amylose content, plastifier type, and content and the storage conditions.

Studies of the barrier properties of starch films are important in order to estimate the shelf life of a food product. This barrier property depends on the starch source, and on the quantity and type of plastifier used, among the most important. Gas barrier properties for a TPS film include water vapor permeability (WVP), oxygen permeability [14], and aroma permeability. WVP is used to describe the ability of the film to control water vapor transportation between a food system and its surrounding. TPS films are not considered good water vapor barriers [68]. The use of CNF has shown to be a viable option for the improvement of mechanical and barrier properties of TPS films. **Table 3** presents a number of studies on the use of starch from different botanical sources to obtain TPS films reinforced with CNF from different materials. However, there are in fact very few reports relating to the use of CNF as reinforcement materials in TPS films. A number of studies have researched the use of undervalued residues as source material for the procurement of cellulose fibers or CNF such as cassava bagasse, barley husk, and sugarcane bagasse, as shown in **Table 3**.

Type of starch	Type of fiber	Preparation of CNF or isolation of cellulose fiber	Preparation of TPS reinforced with CNF or cellulose fibers	Most important results	Reference
Cassava	Cassava	Acidic hydrolysis with	A mixture containing starch,	The addition of 10%	[49]
	bagasse	sulfuric acid $(H_2SO_4)$ at	glycerol or a mixture of glycerol	and 20% of CNF	[]
	euguose	$60^{\circ}$ C for 40 min. Excess	with sorbitol, stearic acid and	significantly reduced	1
		acid was removed by	different quantities of CNF from	the elastic module of	
		centrifugation. Dialysis	cassava bagasse. The films were	the TPS films	
		of the suspension and	prepared by compression		
		ultrasonic treatment	molding at 140°C		
Barley gra	<b>in</b> Barley husk	Removal of lipids from	The films were prepared	The addition of	[72]
20009 800	in buriey music	the barley husk.	by <i>casting</i> . A suspension	cellulose fibers in the	
		Removal of lignin and	containing 3% starch in distilled	films increased the	
		hemicellulose by	water, 0.30 g of glycerol/g dry	TS and decreased	
		alkaline treatment at	starch, 0.01 g of guar gum/g dry	elongation.	
		80°C for 4 h. Bleaching	starch, 10 and 20 g of cellulose	The WVP of the	
		to remove residual	fiber/100 g dry starch.The	starch film with 20%	
			solution was heated at 90°C for	of cellulose fibers	
		buffer and a solution of	10 min and poured onto acrylic	was lower than that	
				of the film without	
		sodium chlorite at 95°C for 4 h	plaques	fibers	
Potato	Wood flour	To obtain the cellulose,	A mixture of starch, sorbitol,	An increase in TS of	[74]
		the wood flour was	stearic acid and CNF (5, 10, 15,	TPS films was	
		treated with acetic acid	and 20 g/100 g dry starch). The	observed with the	
		and sodium chlorite	TPS films with CNF were	addition of CNF	
		between 70 and 75°C	manufactured using a twin		
		for 58 h.The CNF	screw extruder. Pieces of the		
		were obtained from	extruded materials were		
		delignified wood flour	compression molded into thin		
		through mechanical	films with a thickness of 0.3 mm		
		fibrillation			

Type of starch	Type of fiber	Preparation of CNF or isolation of cellulose fiber	Preparation of TPS reinforced with CNF or cellulose fibers	Most important results	Reference
Cassava	Wood cellulose fibers	Not reported	The films were prepared by casting. Film-forming solutions were prepared with 3% w/w of cassava starch, 0.30 g glycerol/g dry starch, 0.01 g guar gum/g dry starch (to avoid fiber sedimentation) and three concentrations of cellulose fibers: 0.10, 0.30, and 0.50 g of fiber/g dry starch. The film forming solution was placed in a Petri dish and dried at 40°C for 16 h	The incorporation of cellulose fibers mechanically reinforces the films, resulting in high TS, low deformation and low WVP	
Corn	Eucalyptus wood fiber	Acid hydrolysis of eucalyptus wood fiber at 50°C for 50 min. The hydrolyzed material was subjected to centrifugation, washing, dialysis, and ultrasonication	The films were prepared by <i>casting</i> . The filmogenic solution was composed of 3% starch, 20% glycerol, gelatin, and CNF. The solution was placed in Petri dishes and dried at 25°C to 50% relative humidity (RH)	The films with gelatin and CNF have greater resistance.	[75]
Chayote	Starch	Acid hydrolysis of the cellulose fiber (H <sub>2</sub> SO <sub>4</sub> , 60%) at 45°C for 30 min	The films were prepared by <i>casting</i> . The filmogenic solution consisted of 4% starch, 2% glycerol, cellulose, or CNF and water. The solution was heated to 90°C for 10 min. The solution was placed in Petri dishes and dried at 40°C for 24 h. The films were placed at 25°C and 57% RH	The addition of CNF reinforced the film matrix, improving TS, EM, and E. The mechanical properties of the starch films with CNF were better than those for films with cellulose. Starch films added with CNF presented the lowest WVP values	
Tamarind seeds	Sugarcane bagasse	Not reported	A mixture of tamarind starch (12%, w/v) and cellulose	The cellulose used	[77]

Type of	Type of	Preparation of CNF or	Preparation of TPS reinforced	Most important Reference
starch	fiber	isolation of cellulose	with CNF or cellulose fibers	results
		fiber		
			(4% w/v) in 100 ml of deionized	as filler improves the
			water. 1.5% of acetic acid and	mechanical
			2.5% of glycerol were added.	properties of the film
			The mixture was gelatinized at	(TS and E)
			105°C for a period of 15–20 min.	
			The mixture was placed on	
			glass trays at 50°C, for 6 h	
Potato	Potato	Potato pulp was treated	A suspension of cellulose	The cellulose [73]
tuber	tuber	with NaOH (2%) at	microfibrils (3.3%) was mixed	microfibrils reinforce
		80°C for 2.5 h. The	with a gelatinized solution	the starch matrix in
		cellulose was	of starch (3.1%). Glycerol was	the film (greater
		submitted to a	used as plasticizer. The mixture	tension module in
		bleaching process with	was homogenized and air	comparison with the
		a solution of sodium	bubbles were eliminated at	film without
		chlorite (NaClO <sub>2</sub> ). The	reduced pressure. The	cellulose
		resultant cellulose was	suspension was poured into a	microfibrils)
		washed with distilled	Teflon mold.	
		water and lyophilized.		
		The cellulose		
		microfibrils were		
		obtained by submitting		
		the cellulose to a		
		homogenization process		
		with distilled water at		
		500 bars and		
		90–95°C		
Corn	_	Extraction of the	The nanocomposites were	The addition of [76]
		crystalline region of	obtained by <i>casting</i> . A solution	waxy starch
		the cellulose was carried	was prepared containing 3.58 g	nanocrystals and
		out by acid hydrolysis	of normal corn starch, 1.93 g	CNF increased the
		(H <sub>2</sub> SO <sub>4</sub> , 64%) for 30 min	glycerol, 35 g of distilled water,	TS and reduced
		at 45°C. The CNF	and different quantities of waxy	WVP of TPS films.
		obtained was washed	corn starch nanocrystals (0, 50,	Moreover, with the
		and neutralized by	and 100%) and CNF (0, 50, and	addition of the CNF,
		dialysis. Finally, the	100%). The solution was	deformation values

Type of	Type of	Preparation of CNF or	Preparation of TPS reinforced	Most important	Reference
starch	fiber	isolation of cellulose	with CNF or cellulose fibers	results	
		fiber			
		suspension was	was placed in Petri dishes and	Young's module	
		sonicated for 15 min	dried at 55°C. The samples	increased	
			were stored at 43% RH for 2		
			weeks		
Wheat	Cotton	The cotton CNF were	A suspension was prepared	As the CNF content	[78]
		obtained by acid	containing 7% wheat starch, 3%	increased,	
		hydrolysis with $H_2SO_4$	glycerol, and 90% water. The	deformation	
		(64%) at 45°C for 4 h.	suspension was gelatinized at	diminished and both	n
		After hydrolysis, the	100°C for 20 min. After	Young's module and	l
		suspension was	gelatinization, a dispersion of	TS of TPS films	
		neutralized and	CNF (0, 2.5, 5, 10, 15, 20, 25, and	increased	
		washed by dialysis	30%) was added to the		
			suspension and mixed for 20 min		
			The mixture was degassed in a		
			vacuum and placed in a		
			polystyrene mold. The		
			nanocomposites were dried		
			at 40°C with 50% RH		

Table 3. Research work on the use of starch from different botanical sources to obtain TPS films reinforced with CNF.

**Table 3** presents information regarding the isolation of cellulose fibers, CNF preparation, the preparation of TPS films reinforced with CNF or with cellulose fibers, and indicates the most significant results.

In general, one can observe that the isolation of cellulose fiber consists in exposing the plant material to high temperatures by means of an alkaline treatment, with the purpose of eliminating lignin and hemicellulose. In addition, the material is exposed to a bleaching process at high temperatures with a solution of sodium chlorite.

For the procurement of CNF, the cellulose fibers are treated with hydrolysis between 45 and 60°C. After hydrolysis, the CNF are recovered by centrifugation, dialysis, and subsequent treatment with ultrasonic bath.

Diverse studies are available which report the use of the casting technique for the production of biodegradable starch films. The most commonly used plasticizer is glycerol. To prepare the solution used to form the films, the starch is mixed with glycerol, water, and different quantities of CNF (between 2.5 and 50% with respect to the starch) as reinforcing agents. In some cases, guar gum is used to avoid sedimentation of the fibers. The solution for film formation is heated in order to achieve gelatinization of the starch for a specific period of time. After gelatinization,

the solution is poured into Petri dishes for the formation of the films, and conditioned for evaluation. Very few studies have been published reporting on the use of injection molding and extrusion to obtain TPS films reinforced with CNF.

The mechanical properties and water barrier properties of TPS films reinforced with CNF have been reported in a number of publications. In general, CNF facilitates an increase in tensile strength, a decrease in deformation values, an increase in Young's module, and a decrease in WVP of TPS films. The chemical structure of cellulose and starch is similar. When they are mixed to produce a filmogenic solution, interactions among the OH groups of both polymers are produced by hydrogen bridges, producing a rigid network that increases the TS [4, 5, 69]. The addition of CNF favors high values of TS for TPS films. This may be due to the fact that a greater contact surface is produced between CNF and the starch chains [70].

The nanometric size of CNF allows a low WVP value of the starch films, which favors the generation of a network of hydrogen bridges between the starch chains and the CNF, causing the water molecule to follow a path with many "curves and bends" and thus reducing its diffusion through the starch films [71]. In addition, the cellulose is less hydrophilic than starch, due to its higher crystallinity and compact microfibrillar arrangement, making it more hydrophobic [69].

**Table 3** shows research works on the use of starch from different botanical sources, such as cassava [4, 49], barley grain [72], potato [73, 74], corn [75, 76], chayote [48], tamarind seeds [77], and wheat [78], to obtain TPS films reinforced with CNF.

## 5. Conclusions

The knowledge that the ecosystem is already considerably contaminated as a consequence of the use of synthetic polymers derived from petroleum, environmental initiatives have now been put in place to promote research work on new products which will be compatible with the environment. The development of biocomposites of TPS with cellulose can reduce the dependency on oil reserves.

In recent times, science and technology has centered on the use of more environmentally friendly, raw materials; emphasizing the importance of the advances in such research on "green" composite materials and cellulose fibers for use in the industry.

It is important to mention that cellulose on a nanometric scale is used as a biodegradable reinforced material to improve the mechanical properties as well as the water barrier properties of the TPS.

According to reports in the literature, the technique for the preparation of cellulose fibers from different materials is very similar, whereas the technique for the preparation of TPS reinforced with CNF presents some differences depending on the type of material processed.

Starches from different sources which do not include corn and potato are used in the development of the green composite material, while the cellulose fibers are obtained from waste material produced by the industries, the aim being to attain the sustainability of natural resources.

Extrusion is one of the most promising methods for processing "green" composite materials and cellulose fibers (TPS-cellulose); however, very little information is available with respect to this.

Based on information found in the literature, it is reported that the incorporation of CNF mechanically reinforces the films, which display high tensile strength, low deformation, and low WVP.

Within the next few years, it is highly likely that the production of biocomposites of TPS filmscellulose will intensify with techniques that are normally used in synthetic polymer processing.

## Author details

Emilio Pérez-Pacheco<sup>1\*</sup>, Jorge Carlos Canto-Pinto<sup>1</sup>, Víctor Manuel Moo-Huchin<sup>1</sup>, Iván Alfredo Estrada-Mota<sup>1</sup>, Raciel Javier Estrada-León<sup>1</sup> and Luis Chel-Guerrero<sup>2</sup>

\*Address all correspondence to: eperez@itescam.edu.mx

1 Cuerpo Académico Bioprocesos, Instituto Tecnológico Superior de Calkiní, en el Estado de Campeche Av. Ah Canul SN por Carretera Federal, Campeche, Mexico

2 Facultad de Ingeniería Química, Universidad Autónoma de Yucatán, Colonia Chuburná de Hidalgo Inn, Mérida, Yucatán, Mexico

## References

- [1] Carvalho AJ. Starch: major sources, properties and applications as thermoplastic materials. Elsevier, Amsterdam; 2008.
- [2] García NL, Ribba L, Dufresne A, Aranguren M, Goyanes S. Effect of glycerol on the morphology of nanocomposites made from thermoplastic starch and starch nanocrystals. Carbohydrate Polymers. 2011;84(1):203–10.
- [3] Yan Q, Hou H, Guo P, Dong H. Effects of extrusion and glycerol content on properties of oxidized and acetylated corn starch-based films. Carbohydrate Polymers. 2012;87(1): 707–12.
- [4] Müller CM, Laurindo JB, Yamashita F. Effect of cellulose fibers on the crystallinity and mechanical properties of starch-based films at different relative humidity values. Carbohydrate Polymers. 2009;77(2):293–9.

- [5] Gaspar M, Benko Z, Dogossy G, Reczey K, Czigany T. Reducing water absorption in compostable starch-based plastics. Polymer Degradation and Stability. 2005;90(3): 563–9.
- [6] Müller CM, Yamashita F, Laurindo JB. Evaluation of the effects of glycerol and sorbitol concentration and water activity on the water barrier properties of cassava starch films through a solubility approach. Carbohydrate Polymers. 2008;72(1):82–7.
- [7] Tashiro K, Kobayashi M. Theoretical evaluation of three-dimensional elastic constants of native and regenerated celluloses: role of hydrogen bonds. Polymer. 1991;32(8):1516–26.
- [8] Peelman N, Ragaert P, De Meulenaer B, Adons D, Peeters R, Cardon L, et al. Application of bioplastics for food packaging. Trends in Food Science & Technology. 2013;32(2):128–41.
- [9] Rincón AM, Rached LB, Aragoza LE, Padilla F. Efecto de la acetilación y oxidación sobre algunas propiedades del almidón de semillas de Fruto de pan (*Artocarpus altilis*). Archivos Latinoamericanos de Nutrición. 2007;57(3):287.
- [10] Matzinos P, Bikiaris D, Kokkou S, Panayiotou C. Processing and characterization of LDPE/starch products. Journal of Applied Polymer Science. 2001;79(14):2548–57.
- [11] Whistler RL, BeMiller JN. Carbohydrate chemistry for food scientists. Eagan Press, St. Paul, Minn. (USA); 1997.
- [12] Fritz H-G, Schroeter J. Study on production of thermoplastics and fibres based mainly on biological materials: European Commission, Directorate-General XII, Science, Research and Development, Agro-Industrial Research Division; 1994.
- [13] Tharanathan RN. Starch—value addition by modification. Critical Reviews in Food Science and Nutrition. 2005;45(5):371–84.
- [14] Manek RV, Kunle OO, Emeje MO, Builders P, Rao GVR, Lopez GP, et al. Physical, thermal and sorption profile of starch obtained from *Tacca leontopetaloides*. Starch-Stärke. 2005;57(2):55–61.
- [15] Cruz BR, Abraão AS, Lemos AM, Nunes FM. Chemical composition and functional properties of native chestnut starch (Castanea sativa Mill). Carbohydrate Polymers. 2013;94(1):594–602.
- [16] Hung PV, Maeda T, Morita N. Study on physicochemical characteristics of waxy and high-amylose wheat starches in comparison with normal wheat starch. Starch-Stärke. 2007;59(3–4):125–31.
- [17] Pérez-Pacheco E, Moo-Huchin V, Estrada-León R, Ortiz-Fernández A, May-Hernández L, Ríos-Soberanis C, et al. Isolation and characterization of starch obtained from *Brosimum alicastrum* Swarts Seeds. Carbohydrate Polymers. 2014;101:920–7.

- [18] Jiménez-Hernández J, Salazar-Montoya J, Ramos-Ramírez E. Physical, chemical and microscopic characterization of a new starch from chayote (*Sechium edule*) tuber and its comparison with potato and maize starches. Carbohydrate Polymers. 2007;68(4):679–86.
- [19] Estrada-León R, Moo-Huchin V, Ríos-Soberanis C, Betancur-Ancona D, May-Hernández L, Carrillo-Sánchez F, et al. The effect of isolation method on properties of parota (*Enterolobium cyclocarpum*) starch. Food Hydrocolloids. 2016;57:1–9.
- [20] Hernandez-Medina M, Torruco-Uco JG, Chel-Guerrero L, Betancur-Ancona D. Physicochemical characterization of starch tuber grown in Yucatan, Mexico. Ciencia e Tecnologia de Alimentos. 2008;28(3):718–26.
- [21] Singh J, Dartois A, Kaur L. Starch digestibility in food matrix: a review. Trends in Food Science & Technology. 2010;21(4):168–80.
- [22] Kaur L, Singh N, Singh J. Factors influencing the properties of hydroxypropylated potato starches. Carbohydrate Polymers. 2004;55(2):211–23.
- [23] González-Reyes E, Méndez-Montealvo G, Solorza-Feria J, Toro-Vazquez J, Bello-Perez LA. Rheological and thermal characterization of Okenia hypogaea (Schlech. & Cham.) starch. Carbohydrate Polymers. 2003;52(3):297–310.
- [24] Ma X, Yu J. The plastcizers containing amide groups for thermoplastic starch. Carbohydrate Polymers. 2004;57(2):197–203.
- [25] Bastioli C. Global status of the production of biobased packaging materials. Starch-Stärke. 2001;53(8):351–5.
- [26] Liu H, Xie F, Yu L, Chen L, Li L. Thermal processing of starch-based polymers. Progress in Polymer Science. 2009;34(12):1348–68.
- [27] Dean K, Yu L, Wu DY. Preparation and characterization of melt-extruded thermoplastic starch/clay nanocomposites. Composites Science and Technology. 2007;67(3):413–21.
- [28] Walenta E, Fink HP, Weigel P, Ganster J, Schaaf E. Structure-property relationships of extruded starch, 2 extrusion products from native starch. Macromolecular Materials and Engineering. 2001;286(8):462–71.
- [29] Chinnaswamy R. Basis of cereal starch expansion. Carbohydrate Polymers. 1993;21(2–3):157–67.
- [30] Guan J, Eskridge KM, Hanna MA. Acetylated starch-polylactic acid loose-fill packaging materials. Industrial Crops and Products. 2005;22(2):109–23.
- [31] Xu Y, Kim KM, Hanna MA, Nag D. Chitosan–starch composite film: preparation and characterization. Industrial crops and Products. 2005;21(2):185–92.

- [32] Lawton, J. W., in: Campbell, G. M., Webb, C., McKee, S. L. (Eds.), Cereals: Novel Uses and Processes, Springer US, Boston, MA; 1997, pp. 43-47.
- [33] Mali S, Grossmann MVE, Garciá, Martino MN, Zaritzky NE. Mechanical and thermal properties of yam starch films. Food Hydrocolloids. 2005;19(1):157–64.
- [34] Shen Y-d, Lai X-j. Structure and mechanical properties of polyethylene glycol modified poly (lactic acid)/thermoplastic starch blend films. Modern Chemical Industry. 2006;26(5):35.
- [35] Averous L, Boquillon N. Biocomposites based on plasticized starch: thermal and mechanical behaviours. Carbohydrate Polymers. 2004;56(2):111–22.
- [36] Bangyekan C, Aht-Ong D, Srikulkit K. Preparation and properties evaluation of chitosan-coated cassava starch films. Carbohydrate Polymers. 2006;63(1):61– 71.
- [37] Bastioli C. Properties and applications of Mater-Bi starch-based materials. Polymer Degradation and Stability. 1998;59(1):263–72.
- [38] Ma X, Yu J, Kennedy JF. Studies on the properties of natural fibers-reinforced thermoplastic starch composites. Carbohydrate Polymers. 2005;62(1):19–24.
- [39] Wuestenberg, T., Cellulose and Cellulose Derivatives in the Food Industry: Fundamentals and Applications, Wiley, Weinheim, Germany; 2014.
- [40] Stana-Kleinschek K, Ribitsch V, Kreze T, Fras L. Determination of the adsorption character of cellulose fibres using surface tension and surface charge. Materials Research Innovations. 2002;6(1):13–8.
- [41] Zugenmaier P. Crystalline cellulose and derivatives: characterization and structures. 1st ed. Springer-Verlag, Berlin Heidelberg; 2008. 285 p.
- [42] Herrick FW, Casebier RL, Hamilton JK, Sandberg KR, editors. Microfibrillated cellulose: morphology and accessibility. J Appl Polym Sci: Appl Polym Symp (United States); 1983: ITT Rayonier Inc., Shelton, WA.
- [43] Turbak AF, Snyder FW, Sandberg KR, editors. Microfibrillated cellulose, a new cellulose product: properties, uses, and commercial potential. J Appl Polym Sci: Appl Polym Symp (United States); 1983: ITT Rayonier Inc., Shelton, WA.
- [44] Arroyo RK., Biocompósitos de Almidón Termoplástico con Microfibras de Celulosa. Instituto Politécnico Nacional, Altamira, Tamaulipas, México. 2008, p. 77.
- [45] Khalil HA, Bhat A, Yusra AI. Green composites from sustainable cellulose nanofibrils: a review. Carbohydrate Polymers. 2012;87(2):963–79.
- [46] de Souza Lima MM, Borsali R. Static and dynamic light scattering from polyelectrolyte microcrystal cellulose. Langmuir. 2002;18(4):992–6.

- [47] Dong XM, Kimura T, Revol J-F, Gray DG. Effects of ionic strength on the isotropic-chiral nematic phase transition of suspensions of cellulose crystallites. Langmuir. 1996;12(8): 2076–82.
- [48] Aila-Suárez S, Palma-Rodríguez HM, Rodríguez-Hernández AI, Hernández-Uribe JP, Bello-Pérez LA, Vargas-Torres A. Characterization of films made with chayote tuber and potato starches blending with cellulose nanoparticles. Carbohydrate Polymers. 2013;98(1):102–7.
- [49] Teixeira EdM, Pasquini D, Curvelo AA, Corradini E, Belgacem MN, Dufresne A. Cassava bagasse cellulose nanofibrils reinforced thermoplastic cassava starch. Carbohydrate Polymers. 2009;78(3):422–31.
- [50] Cranston ED, Gray DG. Morphological and optical characterization of polyelectrolyte multilayers incorporating nanocrystalline cellulose. Biomacromolecules. 2006;7(9): 2522–30.
- [51] Fleming K, Gray D, Prasannan S, Matthews S. Cellulose crystallites: a new and robust liquid crystalline medium for the measurement of residual dipolar couplings. Journal of the American Chemical Society. 2000;122(21):5224–5.
- [52] Battista O.A. Microcrystal polymer science. McGraw-Hill, New York; 1975.
- [53] Roman M, Winter WT. Effect of sulfate groups from sulfuric acid hydrolysis on the thermal degradation behavior of bacterial cellulose. Biomacromolecules. 2004;5(5):1671–7.
- [54] Saito T, Nishiyama Y, Putaux J-L, Vignon M, Isogai A. Homogeneous suspensions of individualized microfibrils from TEMPO-catalyzed oxidation of native cellulose. Biomacromolecules. 2006;7(6):1687–91.
- [55] Pääkkö M, Ankerfors M, Kosonen H, Nykänen A, Ahola S, Österberg M, et al. Enzymatic hydrolysis combined with mechanical shearing and high-pressure homogenization for nanoscale cellulose fibrils and strong gels. Biomacromolecules. 2007;8(6):1934–41.
- [56] Zimmermann T, Pöhler E, Geiger T. Cellulose fibrils for polymer reinforcement. Advanced Engineering Materials. 2004;6(9):754–61.
- [57] Chakraborty A, Sain M, Kortschot M. Cellulose microfibrils: a novel method of preparation using high shear refining and cryocrushing. Holzforschung. 2005;59(1):102–7.
- [58] Zhao H, Kwak JH, Zhang ZC, Brown HM, Arey BW, Holladay JE. Studying cellulose fiber structure by SEM, XRD, NMR and acid hydrolysis. Carbohydrate Polymers. 2007;68(2):235–41.
- [59] Olsson RT, Fogelström L, Martínez-Sanz M, Henriksson M. Cellulose nanofillers for food packaging. Multifunctional and Nanoreinforced Polymers for Food Packaging. 2011 ed. Cambridge: Woodhead Publishing Limited; 2011. pp. 86–107.

- [60] Abe K, Yano H. Comparison of the characteristics of cellulose microfibril aggregates of wood, rice straw and potato tuber. Cellulose. 2009;16(6):1017–23.
- [61] Abe K, Yano H. Comparison of the characteristics of cellulose microfibril aggregates isolated from fiber and parenchyma cells of Moso bamboo (*Phyllostachys pubescens*). Cellulose. 2010;17(2):271–7.
- [62] Oksman K, Mathew A, Bondeson D, Kvien I. Manufacturing process of cellulose whiskers/polylactic acid nanocomposites. Composites Science and Technology. 2006;66(15):2776–84.
- [63] Sorrentino A, Gorrasi G, Vittoria V. Potential perspectives of bio-nanocomposites for food packaging applications. Trends in Food Science & Technology. 2007;18(2):84–95.
- [64] Raj R, Kokta B, Dembele F, Sanschagrain B. Compounding of cellulose fibers with polypropylene: effect of fiber treatment on dispersion in the polymer matirx. Journal of Applied Polymer Science. 1989;38(11):1987–96.
- [65] Janssen, L., Moscicki, L., Thermoplastic starch, John Wiley & Sons, Weinheim, Germany; 2009.
- [66] Wattanakornsiri A, Tongnunui S. Sustainable green composites of thermoplastic starch and cellulose fibers. Songklanakarin Journal of Science and Technology. 2014;36(2):149– 61.
- [67] Mali S, Sakanaka L, Yamashita F, Grossmann M. Water sorption and mechanical properties of cassava starch films and their relation to plasticizing effect. Carbohydrate Polymers. 2005;60(3):283–9.
- [68] Zhang Y, Rempel C, Liu Q. Thermoplastic starch processing and characteristics—a review. Critical Reviews in Food Science and Nutrition. 2014;54(10):1353–70.
- [69] Curvelo A, De Carvalho A, Agnelli J. Thermoplastic starch–cellulosic fibers composites: preliminary results. Carbohydrate Polymers. 2001;45(2):183–8.
- [70] Teixeira EdM, Lotti C, Corrêa AC, Teodoro KB, Marconcini JM, Mattoso LH. Thermoplastic corn starch reinforced with cotton cellulose nanofibers. Journal of Applied Polymer Science. 2011;120(4):2428–33.
- [71] De Azeredo HM. Nanocomposites for food packaging applications. Food Research International. 2009;42(9):1240–53.
- [72] El Halal SLM, Colussi R, Deon VG, Pinto VZ, Villanova FA, Carreño NLV, et al. Films based on oxidized starch and cellulose from barley. Carbohydrate Polymers. 2015;133:644–53.
- [73] Dufresne A, Dupeyre D, Vignon MR. Cellulose microfibrils from potato tuber cells: processing and characterization of starch–cellulose microfibril composites. Journal of Applied Polymer Science. 2000;76(14):2080–92.

- [74] Hietala M, Mathew AP, Oksman K. Bionanocomposites of thermoplastic starch and cellulose nanofibers manufactured using twin-screw extrusion. European Polymer Journal. 2013;49(4):950–6.
- [75] Freitas F, Alves VD, Reis MA, Crespo JG, Coelhoso IM. Microbial polysaccharide-based membranes: current and future applications. Journal of Applied Polymer Science. 2014;131(6). doi: 10.1002/app.40047
- [76] Gonzalez JS, Ludueña LN, Ponce A, Alvarez VA. Poly(vinyl alcohol)/cellulose nanowhiskers nanocomposite hydrogels for potential wound dressings. Materials Science and Engineering: C. 2014;34:54–61.
- [77] Sudharsan K, Mohan CC, Babu PAS, Archana G, Sabina K, Sivarajan M, et al. Production and characterization of cellulose reinforced starch (CRT) films. International Journal of Biological Macromolecules. 2016;83:385–95.
- [78] Lu Y, Weng L, Cao X. Biocomposites of plasticized starch reinforced with cellulose crystallites from cottonseed linter. Macromolecular Bioscience. 2005;5(11):1101–7.





IntechOpen