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# Comparative Study of Materials to SF<sub>6</sub> Decomposition Components

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<http://dx.doi.org/10.5772/intechopen.68322>

## Abstract

In order to judge the inside insulation fault of SF<sub>6</sub> insulated equipment, the gas-sensing properties to a series of characteristic SF<sub>6</sub> decomposition components, SOF<sub>2</sub>, SO<sub>2</sub>F<sub>2</sub>, SO<sub>2</sub>, H<sub>2</sub>S, CF<sub>4</sub>, HF, and SF<sub>6</sub>, have been studied. In this study, a comparative study of these gas-sensing materials has been made in theoretical and experimental fields to find the optimal gas-sensing material, and put forward the effective approaches to improve the gas-sensing properties of materials.

**Keywords:** SF<sub>6</sub> decomposed gases, carbon nanotubes, TiO<sub>2</sub> nanotubes, graphene gas sensor, gas-sensing comparison

## 1. Theoretical study comparison

To detect, evaluate, and diagnose the insulation status of SF<sub>6</sub> gas insulated equipment using the characteristics of SF<sub>6</sub> decomposition, a series of characteristic SF<sub>6</sub> decomposition components, SOF<sub>2</sub>, SO<sub>2</sub>F<sub>2</sub>, SO<sub>2</sub>, H<sub>2</sub>S, CF<sub>4</sub>, HF, and SF<sub>6</sub>, are detected by gas sensors, including metal functionalized single wall carbon nanotubes (SWCNTs) [1–6], TiO<sub>2</sub> nanotubes [7–9], and graphene gas sensors [10, 11], as shown in **Table 1**. According to the theoretical calculation results, these three kinds of gas-sensing materials are not sensitive to the background gas SF<sub>6</sub> and decomposition component CF<sub>4</sub>. And the concentration of HF in decomposition components is less than the concentration of the other typical gases. Therefore, we compare the adsorption properties four types of SF<sub>6</sub> decomposition components: SOF<sub>2</sub>, SO<sub>2</sub>F<sub>2</sub>, SO<sub>2</sub>, and H<sub>2</sub>S, under the detection of metal functionalized SWCNTs, TiO<sub>2</sub> nanotubes, and graphene gas sensors.

Material	Dopant	Typical decomposition components						
SWCNTs	/	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	H <sub>2</sub> S	CF <sub>4</sub>	HF	SF <sub>6</sub>
	Pd	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	H <sub>2</sub> S	CF <sub>4</sub>	/	/
	Ni	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	/	/	/	/
	Al	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	H <sub>2</sub> S	CF <sub>4</sub>	/	/
	Pt	/	/	SO <sub>2</sub>	H <sub>2</sub> S	/	/	/
	Au	/	/	SO <sub>2</sub>	H <sub>2</sub> S	/	/	/
TiO <sub>2</sub>	/	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	/	/	/	/
	Pt	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	/	/	/	/
	Au	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	/	/	/	/
Graphene	/	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	H <sub>2</sub> S	/	/	/
	Au	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>	SO <sub>2</sub>	H <sub>2</sub> S	/	/	/

**Table 1.** The typical decomposition components of SF<sub>6</sub> detected by gas sensor.

Metal atom modification not only greatly improves the gas-sensing properties to SF<sub>6</sub> decomposition components, but also enhances the gas-sensing selectivity of sensors to different gas molecules. Upon SOF<sub>2</sub> and SO<sub>2</sub>F<sub>2</sub> detection, intrinsic gas sensors show weak sensitivity to SOF<sub>2</sub> because of the weak interaction between gas molecules and the surface of gas-sensing materials. After metal atoms modification, which acts as the active sites, SOF<sub>2</sub> molecule tends to approaches the adsorption site by fluorine atom due the its strong chemical activity. Generally, the strong activity breaks the chemical bonds of SOF<sub>2</sub>. Similarly, SO<sub>2</sub>F<sub>2</sub> interacts with the metal atom-doped gas-sensing material by chemisorption, reflecting in aspect that the fluorine atom breaks from SO<sub>2</sub>F<sub>2</sub> to build new bond with metal atoms. Upon SO<sub>2</sub> and H<sub>2</sub>S detection, these two small gas molecules are generally adsorbed on the surface of gas sensors by physisorption, which was benefit to gas desorption process. SO<sub>2</sub> and H<sub>2</sub>S get approach to the surrounding of metal dopant by oxygen and sulfur atom, respectively, because of its polyvalency property.

Comparing the gas-sensing properties of different sensor materials, the regular porous structure of TiO<sub>2</sub> nanotubes contributes to gas desorption and reusability. In addition, its big specific surface area helps the metal dopant modification and gas-sensing sites. But, TiO<sub>2</sub> nanotubes sensor usually needs high work temperature, and its high resistance hinders the transmission of detection signals. In comparison, metal atom-doped SWCNT and graphene sensor can work effectively at room temperature, and its low resistance helps to transfer detection signals, resulting in reducing gas-sensing time.

2. Experimental study comparison

Using the own design platform which is used to test the performance of TiO<sub>2</sub> nanotubes sensor, the gas response characteristics and temperature characteristics of the intrinsic TiO<sub>2</sub> nanotubes sensor to three main SF<sub>6</sub> gas decomposition compositions SO<sub>2</sub>, SOF<sub>2</sub>, and SO<sub>2</sub>F<sub>2</sub> were studied. The same sensing experiments were also carried out on Pt and Au-doped TiO<sub>2</sub>

nanotubes sensors. The results are compared as shown in **Table 2** for TiO<sub>2</sub> nanotubes gas sensor and **Table 3** for graphene gas sensor.

It is concluded that the Au-doped TiO<sub>2</sub> nanotubes sensor has better selectivity to SO<sub>2</sub>F<sub>2</sub> gas. Pt nanoparticles doping changes gas selectivity of TiO<sub>2</sub> nanotubes sensor to SO<sub>2</sub>, SOF<sub>2</sub>, and SO<sub>2</sub>F<sub>2</sub>. Compared with the intrinsic sensor, Au nanoparticles doped significantly changed the selectivity of sensor to SO<sub>2</sub>, SOF<sub>2</sub>, and SO<sub>2</sub>F<sub>2</sub> (**Table 3**).

Pristine graphene is considered a promising adsorbent for H<sub>2</sub>S selective detection. Compared with the performance on pristine graphene films, Au-doped graphene emerges significant response to H<sub>2</sub>S, SOF<sub>2</sub>, and SO<sub>2</sub>F<sub>2</sub> but weak interaction to SO<sub>2</sub>, with the sequence of SO<sub>2</sub>F<sub>2</sub> > H<sub>2</sub>S > SOF<sub>2</sub> > SO<sub>2</sub>. Among them, only H<sub>2</sub>S shows the opposite response with its resistance increase, while SO<sub>2</sub>, SOF<sub>2</sub>, and SO<sub>2</sub>F<sub>2</sub> decrease the resistance of Au-doped graphene.

Doping metal	Doping time	Gas-sensing parameters to typical decomposition components		
		SO <sub>2</sub>	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>
/	0 s	-74.6%	-7.82%	-5.52%
Pt	10 s	-53.3%	-24.2%	-5.4%
	20 s	-33.9%	-22.1%	-19.2%
	30 s	-13.8%	-19.1%	-50.6%
	40 s	-6.7%	-5.1%	-10.7%
Au	–	-23.75%	-28.37%	-42.31%

**Table 2.** The typical decomposition components of SF<sub>6</sub> detected by TiO<sub>2</sub> nanotubes gas sensor.

Doping metal	Gas-sensing parameters to typical decomposition components			
	H <sub>2</sub> S	SO <sub>2</sub>	SOF <sub>2</sub>	SO <sub>2</sub> F <sub>2</sub>
/	-15.78%	-1.5%	-0.8%	-0.3%
Au	28.15%	-6.98%	-23.83%	-33.91%

**Table 3.** The typical decomposition components of SF<sub>6</sub> detected by graphene gas sensor.

### 3. Conclusions

In order to evaluate and diagnose the insulation status of SF<sub>6</sub> insulated equipment, gas sensor detection becomes an effective new method to realize the function by detecting the decomposition components of SF<sub>6</sub>. Theoretical simulations are performed to understand the adsorption process of gas sensors and typical components of SF<sub>6</sub>. And using carbon nanotubes (CNTs) based as novel kind of sensors show high sensitivity and quick responses to target gases. For TiO<sub>2</sub>-based gas sensor, the adsorption of typical components of SF<sub>6</sub> on different surfaces of TiO<sub>2</sub> is reviewed in this section. It is found that the metal decoration improves the sensitivity

and selectivity to SO<sub>2</sub> and SOF<sub>2</sub>, and SO<sub>2</sub>F<sub>2</sub> also reduces the working temperature for gas detection. Pristine graphene exhibits weak adsorption and absence of charge transfer, which indicates barely satisfactory sensing for decomposed components. SOF<sub>2</sub> and SO<sub>2</sub>F<sub>2</sub> exhibit a strong chemisorption interaction with Au-graphene, while H<sub>2</sub>S and SO<sub>2</sub> exhibit quasi-molecular binding effects. Only H<sub>2</sub>S exhibits n-type doping to Au-graphene, whereas the rest gases exhibit p-type doping. In general, the sensors array composed of modified gas sensors can be used in the GIS to realize the highly precise detection of related gases, thus accurately deducing the related insulation faults.

## Author details

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