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### Effect of Buoyancy on Pore-Scale Characteristics of Two-Phase Flow in Porous Media

Tetsuya Suekane and Hiroki Ushita The University of Tokushima Japan

#### 1. Introduction

Carbon dioxide (CO<sub>2</sub>) is the most important anthropogenic greenhouse gas. The global atmospheric concentration of CO<sub>2</sub> has increased from a pre-industrial value of approximately 280 ppm to 379 ppm in 2005. The warming of the climate system is unequivocal, as is now evident from observations of increasing global average air and ocean temperatures, widespread melting of snow and ice and rising global average sea level (IPCC, 2007a). To stabilize the concentration of CO<sub>2</sub> in the atmosphere, emissions need to peak and then decline thereafter. In the long term, energy conservation, efficiency improvements in energy conversion, lower carbon fuels such as natural gas and renewable energy sources are the most promising alternatives. For lower stabilization targets, scenarios put more emphasis on the use of  $CO_2$  capture and storage (CCS; Pacala & Socolow, 2004; IPCC, 2007b); however, the transition from the current dependence on fossil fuels would take many decades. The capture of  $CO_2$  from fossil fuel power plants and other large-scale stationary emission sources and storage in geologic formations is the only option that permits a transition from current high-intensity carbon-based energy sources to low-carbon energy sources.

The safety of geologic storage of  $CO_2$  is obviously a central concern in planning carbon sequestration on a large scale. The current concept of geologic storage involves the injection of  $CO_2$  into deep formations, which typically contain brine.  $CO_2$  is supercritical at temperatures and pressures above the critical values of 304 K and 7.38 MPa. In typical geologic formations, the critical condition of  $CO_2$  is reached at a depth of approximately 740 m. Because of geologic pressure, the density of  $CO_2$  dramatically increases with depth; however, the density of  $CO_2$  is approximately 0.9 times that of water, so when  $CO_2$  is injected into the subsurface, buoyancy tends to bring  $CO_2$  upward in geologic formations. On the other hand,  $CO_2$  will be retained by physical and geochemical mechanisms, such as physical trapping (IPCC 2005), capillary trapping (Suekane et al. 2008, 2010a; Al Mansoori et al. 2010; Pentland et al. 2010; Zhou et al. 2010; Wildenschild et al. 2010; Saadatpoor 2010), solubility trapping (Lindeberg & Wessel-Berg 1997; McPherson & Cole 2000; Ennis-King et al. 2003; Gilfillan et al. 2009; Iding & Blunt 2010) and mineralization (Gunter et al. 1993).

Capillary trapping is sometimes referred to as residual gas trapping or relative permeability hysteresis trapping. When  $CO_2$  is injected into the subsurface, it spreads in geologic

formations in a continuous phase, displacing brine. As it migrates through a formation, the saturation of  $CO_2$  decreases and some of it is retained in pore space by capillary forces. In the case of Berea sandstone, residual  $CO_2$  saturation ranges from 24.8 to 28.2% at supercritical conditions (Suekane et al. 2008). Once the gas bubbles are trapped, they are stable against water flow, because the capillary forces acting on the gas bubbles are much higher than buoyancy and viscous sheer stress (Suekane et al. 2010a; Zhou et al. 2010).

Saturation of  $CO_2$  trapped by capillarity also depends on numerous factors, such as how the wetting fluid gets in (either by forced or spontaneous imbibition), the rate of imbibition,  $CO_2$  saturation at flow reversal and the properties of porous media (Holtz 2002; Suekane et al. 2010b). In the case of the Sleipner project,  $CO_2$  was injected into a point approximately 200 m below the caprock. The  $CO_2$  plume spreads upward by buoyancy before accumulating below the caprock (IPCC 2005; Arts et al. 2008). On the other hand, in the case of oil production by gas-assisted gravity drainage (GAGD) processes, crestal gas injection uses wells in higher structural positions close to the top of the reservoir. GAGD has been considered to be a very attractive oil recovery process because of its higher efficiency (Hagoort 1980; de Mello et al. 2009; Rostami et al. 2010). When gas is injected vertically downward into the formations, the higher sweep efficiency could be achieved with the aid of stabilization of a displacement front by buoyancy. For upward displacement of fluids by buoyant gas, however, fingering and instability of a displacement front reduces the sweep efficiency and gas saturation. The gas saturation at the flow reversal has strong influence on the residual gas saturation.

In this study, the effect of the stability of a displacing front in gravity drainage on the initial gas saturation is discussed using dimensionless parameters and the relationship between initial gas saturation and residual gas saturation is explored. Gas injection experiments were carried out with packed beds of glass beads using a nitrogen and water system in laboratory conditions. The three-dimensional structure of the distribution of gas in the packed beds was visualized by means of a microfocused X-ray CT scanner. Water was injected into the packed beds to evaluate the residual gas saturation. Finally, gas was injected vertically upward into the packed bed to study the effect of the direction of gas injection with respect to gravity on the gas saturation.

#### 2. Experiments

#### 2.1 Experimental apparatus and procedure

Glass beads with a diameter of 100 µm, 200 µm or 400 µm were packed in an acrylic resin tube with an interior diameter of 10 mm and a height of 40 mm (Fig. 1a). First, the packed bed was soaked in water. After gas contained in the packed bed was evacuated by a vacuum pump, water was forced into pore spaces during the recovery of ambient pressure to atmospheric pressure. Next the packed bed, aligned vertically, was connected by Teflon tube to a gas reservoir containing nitrogen at a pressure of 0.12 MPa and a syringe pump filled with water (Fig. 1b). Nitrogen was injected into the packed bed vertically downward or upward at a constant flow rate, by withdrawing water into the syringe pump at a constant flow rate. Five units of pore volume (PV) of nitrogen were injected into the packed bed. This condition is often referred to as an irreducible water condition, where the remaining water in a porous media is immobile. In this paper, we define the gas saturation at this condition to be the initial gas saturation. Then, the packed bed was disconnected from the tubing systems and placed in an X-ray CT scanner (Comscantechno Co. ScanXmate-RB090SS) to observe the distribution of gas and water. Following the CT scan, the packed bed was connected again to the syringe pump (Fig. 1c). Five PV of water was injected vertically upwards into the packed bed at a constant flow rate. The resulting condition is referred to as the residual gas condition, where gas bubbles are trapped by capillarity in porous media. The saturation at this condition is referred as residual gas saturation. The packed bed was scanned by the X-ray CT scanner again. An X-ray CT scan was performed at the steady state of the packed bed after each gas or water injection.



Fig. 1. Schematic views of (a) the packed bed of glass beads, (b) experimental setup for the drainage process and (c) experimental setup for the imbibition process.

## 2.2 Dimensionless parameters and experimental conditions 2.2.1 Dimensionless parameters

Two-phase flows in porous media are influenced by interfacial tension, buoyancy and viscous sheer stress. The Bond number (Bo) and capillary number (Ca) are defined as

$$Bo = \frac{(\rho_w - \rho_n)gR^2}{\sigma} \tag{1}$$

$$Ca = \frac{\mu_w v}{\sigma}, \qquad (2)$$

where  $\rho$  is the density, g is the acceleration due to gravity, R is the particle radius,  $\sigma$  is interfacial tension,  $\mu$  is the viscosity, v is the displacing fluid velocity and subscripts w and n denote the wetting phase and non-wetting phase, respectively. Bo and Ca represent the ratio of buoyancy force and of viscous sheer stress to the capillary force, respectively. Instead of

Equation 1, the Dombrowski-Brownell number, which is the ratio of the pore scale hydrostatic pressure drop to the capillary pressure (Rostami et al. 2010), is defined as

$$N_{DB} = \frac{(\rho_w - \rho_n)g(k/\phi)}{\sigma},$$
(3)

where k is the absolute permeability and  $\phi$  is the porosity. The semi-heuristic Carman-Kozney model of permeability predicts the permeability of packed beds as follows (Kaviany 1995):

$$k = \frac{\phi^3}{45(1-\phi)^2} R^2 \,. \tag{4}$$

For gravity drainage processes, which correspond to vertically downward injections of gas into the packed bed in the current experiments, the critical gravity drainage velocity is defined by Blackwell and Terry (1959) and Dumore (1964) as

$$v_c = \frac{(\rho_w - \rho_n)gk}{\mu_w} \,. \tag{5}$$

The critical gravity drainage velocity depends on the difference in the density, the viscosity of drained fluid and the permeability of porous media. In the current experiments, because we used the nitrogen and water system in laboratory conditions, the critical gravity drainage velocity depends only on the radius of glass beads from equation 5. The ratio of the critical gravity drainage velocity to the displacing fluid velocity, vc/v, is referred to as the stability parameter.

#### 2.2.2 Experimental conditions

All experiments were performed for the nitrogen and water system in laboratory conditions. The fluid properties of the nitrogen and water system and the supercritical  $CO_2$  and water system at a typical reservoir condition, which corresponds to the depth of approximately 850 m, are summarized in Table 1.

		Viscosity, µ (µPa ·s)	density, ρ (kg/m³)	interfacial tension, σ (mN/m)	Са	Во
Reservoir condition (318 K, 8.5 MPa)	sc CO <sub>2</sub>	20	259.6	35.7	1.681 × 10-2 × v	2.017 × 10 <sup>5</sup> × R <sup>2</sup>
	H <sub>2</sub> O	600	993.9			
Laboratory condition	gas N <sub>2</sub>	17.87	1.123	72.6	1.463 × 10 <sup>-2</sup> × v	1.345 × 10 <sup>5</sup> × R <sup>2</sup>
(293 K, 0.1 MPa)	H <sub>2</sub> O	1062	996.7	72.0		

Table 1 F	Thuid r	properties	under	experimental	conditions and	1 typica	l reservoir	conditions
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We used nitrogen as a non-wetting phase instead of  $CO_2$  for these experiments to reduce the dissolution in water. Interfacial tension between supercritical  $CO_2$  and water is

approximately half of that between nitrogen and water in laboratory conditions. Because of the high density of supercritical  $CO_2$  with respect to nitrogen, buoyancy is also lower for supercritical  $CO_2$ . As a result, the Bond number falls in a similar range.

The water used in the experiments was doped with sodium iodide (NaI) at 7.5 wt% to enhance the X-ray attenuation. Henceforth, for simplicity, this aqueous phase is referred to as water. During drainage processes, the injection flow rate of nitrogen was controlled by a syringe pump to cause the capillary number to be in the range between  $3.72 \times 10^{-7}$  and  $1.86 \times 10^{-5}$ . Depending on the radius of the glass beads, the stability parameter vc/v ranges from 0.05 to 37.9. In imbibition processes, the injection flow rate of water was controlled such that the capillary number was  $1.00 \times 10^{-6}$ .

Reconstructed three-dimensional images are  $608 \times 608 \times 610$  pixels at a resolution of 25.048  $\mu$ m/pixel in all directions.

#### 3. Results and discussion

#### 3.1 Stability effect of gravity drainage on gas saturation

#### 3.1.1 Stable gravity drainage

Nitrogen was injected vertically downward into the packed bed filled with water at various flow rates to investigate the effect of the instability of a displacing front on gas saturation. Figure 2 shows vertical cross-sectional images around the axis of the packed bed. For drainage at stability parameters of 9.2 and 1.8, high initial gas saturations, Sg\*, of 91 and 85%, respectively, are achieved without the effect of gas fingering with the aid of a stable interface of displacement. When the displacement velocity is above the critical gravity drainage velocity, large fractions of water remain in the packed bed. At stability parameters lower than 0.36, fingering of injected gas results in gas saturations below 25%. The distribution of gas shown in Fig. 2 suggests that the critical gravity drainage velocity defined by equation 5 gives an appropriate criterion for the stability of a drainage interface.

#### 3.1.2 Fingering

Figure 3 shows vertical cross-sectional images after the unstable drainage of vc/v = 0.36 for four independent experimental runs under the same conditions, where the capillary number is  $9.30 \times 10^{-6}$  and the Bond number is  $5.38 \times 10^{-3}$ . Because the drainage is unstable, fingering has a great influence on gas distributions. Reflecting the nature of instability, the initial gas saturation varies widely from 22 to 67%. Because the glass beads on the surface of cylindrical tube tend to be sorted, the porosity at the region adjacent to the surface is higher than that at the centre of the packed bed. Figure 3c suggests that heterogeneity in porosity enhances the effect of fingering and reduces the displacement efficiency.

#### 3.1.3 Effect of capillary number on initial gas saturation

The effect of capillary number on the initial gas saturation is shown in Fig. 4 for packed beds of glass beads with the diameters 100  $\mu$ m, 200  $\mu$ m and 400  $\mu$ m. With an increase in capillary number, the initial gas saturation decreases, as has been reported by many researchers (Morrow & Songkran, 1982; Chatzis et al. 1983; Morrow et al. 1988; Rostami et al. 2010). At a capillary number of 9.30 × 10<sup>-6</sup> with 200- $\mu$ m diameter glass beads, the initial gas saturation varies widely due to fingering of the interface (Fig. 3). At a low capillary number of 3.72 × 10<sup>-7</sup>, an initial gas saturation of more than 70% can be achieved, even for the fine glass beads.





Fig. 2. Effect of the stability parameter vc/v on gravity drainage by gas injection. (a) Ca =  $1.86 \times 10^{-7}$ , vc/v = 9.2, Sg\* = 91% (b) Ca =  $3.72 \times 10^{-7}$ , vc/v = 1.8, Sg\* = 85% (c) Ca =  $1.86 \times 10^{-7}$ , vc/v = 1.8, Sg\* =  $1.86 \times 10^{-7}$ , vc/v = 1.8, Sg\* =  $1.86 \times 10^{-7}$ , vc/v = 1.8, Sg\* =  $1.86 \times 10^{-7}$ , vc/v = 1.8 $10^{-6}$ , vc/v = 0.92, Sg\* = 69% (d) Ca =  $9.30 \times 10^{-6}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (e) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 0.36, Sg\* = 25% (f) Ca =  $1.86 \times 10^{-5}$ , vc/v = 1vc/v = 0.18, Sg<sup>\*</sup> = 24%. Glass beads with a diameter of 200  $\mu$ m were packed in a tube with an inner diameter of 10 mm, shown as black regions at the image edge.

#### 3.1.4 Bond number effect on initial gas saturation

The effect of Bond number on the initial gas saturation is shown in Fig. 5 for packed beds of glass beads with various diameters at a constant injection flow rate corresponding to a capillary number of  $3.72 \times 10^{-7}$ . Because the higher Bond number results in a more stabilized displacement front, higher displacement efficiency could be achieved for high Bond numbers (Fig. 4).

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Fig. 3. Reproducibility of gravity drainage in the unstable condition where Ca =  $9.30 \times 10^{-6}$  and Bo =  $5.38 \times 10^{-3}$ .



Fig. 4. Effect of capillary number on the initial gas saturation for drainage processes with downward gas injection.



Fig. 5. Bond number effect on gravity drainage by gas injection at the capillary number of  $3.72 \times 10^{-7}$ . (a) Bo =  $3.36 \times 10^{-4}$ , R = 50 µm (b) Bo =  $1.35 \times 10^{-3}$ , R = 100 µm (c) Bo =  $5.38 \times 10^{-3}$ , R = 200 µm.

The Bond number effect on the initial gas saturation for drainage processes is shown in Fig. 6 for various capillary numbers. With an increase in Bond number, the initial gas saturation increases. With higher diameter glass beads, the higher critical gravity drainage velocity results in the higher displacement efficiency associated with the reduction in capillary force (Rostami et al. 2010). The displacement front of drainage is stable for all experimental runs shown in Fig. 5. Therefore, for the large, 400  $\mu$ m glass beads, an extremely high gas saturation of 98% can be achieved at a high Bond number. This is an attractive fact that suggests high oil recovery of GAGD.



Fig. 6. Bond number effect on the initial gas saturation for drainage processes with downward injection of gas.

## 3.2 Effect of initial gas saturation on residual gas saturation 3.2.1 Morphology of trapped gas bubbles

Water is injected vertically upward into the packed bed of glass beads at a constant flow rate, which corresponds to a capillary number of  $1.0 \times 10^{-6}$ , after gravity drainage processes. Figure 7 shows the distributions of gas after drainage processes and the distributions of trapped gas bubbles after the imbibition processes for each diameter of glass beads. Because of low gas-injection flow rate at a capillary number of  $3.72 \times 10^{-7}$ , high initial gas saturations are achieved for each packed bed after the drainage processes (Fig. 7a-c). Into these packed beds, 5 PV of water was injected vertically upwards at a capillary number of  $1.0 \times 10^{-6}$ . Distributions of trapped gas bubbles are shown in Fig. 7d-f for each diameter of glass beads. As the diameter of glass beads decreases, residual gas saturation increases because the capillary force becomes high compared with buoyancy. Gas is perfectly non-wetting to glass beads. Therefore, gas bubbles with low volume are trapped at the centre of pore spaces. Gas bubbles with a volumetric scale of several pores are trapped in the packed beds spreading over several pores without surface contact of glass beads by thin water films.



(d) (e) (f) Fig. 7. Distributions of injected gas after stable drainage (a-c) and trapped gas bubbles after water imbibition (d-e) for the packed bed of glass beads with a diameter of (a, d) 400  $\mu$ m, (b, e) 200  $\mu$ m and (c, f) 100  $\mu$ m. Surface of the gas in the cylindrical domain with a diameter of 5 mm and a length of 5 mm is visualized.

Distributions of the volume of trapped gas bubbles are shown in Fig. 8. The volume of gas bubbles were analysed using the image processing software ImageJ (Abramoff et al. 2004; Rasband 1997–2008) with some plug-ins for the cylindrical domain at a diameter of 9 mm and length of 2.5 mm. The volume of bubbles is normalized with that of packed glass beads. For lower diameter glass beads, the largest gas bubble tends to be large compared to the glass bead. The volume of the largest bubble in the packed bed of 400  $\mu$ m glass beads is approximately one order of magnitude larger than that of the glass bead. On the other hand, in a packed bed of 100  $\mu$ m glass beads, the volume of the largest gas bubble contains approximately 98% of all trapped gas.



Fig. 8. Distribution of the volume of trapped gas bubbles in packed beds of glass beads with various diameters. The volume of gas bubbles is normalized with that of a glass bead in the packed bed. Vertical axis denotes the volume fraction with respect to the total volume of trapped gas.

#### 3.2.2 Relationship between initial and residual gas saturations

An increase in the injection flow rate of gas results in a decrease in the initial gas saturation (Fig. 4) because of the instability of displacing fronts. Residual gas saturation is affected by initial gas saturation, even for water injection at the same capillary number. The relationship between the initial gas saturation and residual gas saturation is shown in Fig. 9. The residual gas saturation peaks against the initial gas saturation at approximately 50%. An inverse trend between initial gas saturation and residual gas saturation has been found in unconsolidated sandstone (Holtz 2002) and in the packed bed (Suekane et al. 2010b). During imbibition processes, the migration of gas is assisted by buoyancy, because water is injected vertically upward. For higher initial gas saturations, gas in a continuous phase is hardly disconnected from the continuum. Because the displacing front of water retreats due to buoyancy before being disconnected by capillary forces (Setiawan et al. 2010). From a viewpoint of the safety of geologic storage of  $CO_2$ , the difference between initial and residual gas saturations,

which denotes the fraction of  $CO_2$  escaping through a porous media, would be reduced by the design of injection strategy to adjust the initial gas saturation.



Initial gas saturation [%]



#### 3.2.3 Stability of gravity drainage and gas trapping

The initial gas saturation for the packed bed of 400  $\mu$ m glass beads after drainage at stability parameters vc/v of 37.9 and 0.76, was 98 (Fig. 7a) and 79%, respectively. After water imbibition, the residual gas saturation was 3 and 10%, respectively, and the trapped gas bubbles are shown in Figs. 7d and 10, respectively. Distribution of the volume of trapped gas bubbles is shown in Fig. 11. In the case of unstable drainage, even though the initial gas saturation is lower than that in the case of stable drainage, residual gas saturation is high, because large bubbles with the scale of several pore sizes remain in porous media.



Fig. 10. Trapped gas bubbles in the packed bed of 400  $\mu$ m glass beads at a residual gas saturation of 10%, after unstable drainage at a stability parameter vc/v of 0.76.



Normalized volume [-]

Fig. 11. Distribution of the volume of trapped gas bubbles in packed beds of 400  $\mu$ m glass beads. After drainage at stability parameters vc/v of 37.9 and 0.76, water was injected at a capillary number of 1.0 × 10<sup>-6</sup>. The volume of gas bubbles is normalized with that of a glass bead in the packed bed.



Fig. 12. Vertically upward drainage by gas injection at a capillary number of (a) Ca =  $1.86 \times 10^{-7}$ , (b) Ca =  $3.72 \times 10^{-7}$ , (c) Ca =  $1.86 \times 10^{-6}$ , (d) Ca =  $9.30 \times 10^{-6}$ , (e) Ca =  $1.86 \times 10^{-5}$ . Glass beads with a diameter of 200 µm were packed in tube with an inner diameter of 10 mm, shown as black regions at the image edges.

#### 3.3 Upward drainage

In this section, we injected gas vertically upward as is often the case in actual CCS projects. The packed bed of glass beads and experimental setup used in the experiments were the same as that shown in Fig. 1, except for the direction of gas injection.

Figure 12 shows vertical cross-sectional images at various injection flow rates of gas. Except for the direction of gas injection, gas injection flow rates in Fig. 12a–e are the same as those in Fig. 2a–e, respectively. The initial gas saturations are compared for the direction of gas injection in Fig. 13. In the case of the upward injection of gas, gravitational force and capillary pressure let water remain in porous media against gas migration. With a decrease in the gas injection flow rate, the initial gas saturation increases; however, the initial gas saturation at the upward gas injection is lower than that at the downward gas injection at the same capillary number because of instability of displacing front.



Fig. 13. Effect of the direction of gas injection on the initial gas saturation at various capillary numbers.

#### 4. Conclusion

Gas injection experiments were carried out using packed beds of glass beads with a nitrogen and water system in laboratory conditions. The three-dimensional structure of the distribution of gas in the packed beds was visualized by means of a microfocused X-ray CT scanner.

First, the effect of the stability of a displacing front in gravity drainage on the initial gas saturation was discussed with dimensionless parameters. For gravity drainage in stable conditions, high initial gas saturation is achieved without the effect of gas fingering with the aid of a stable interface of displacement. When the displacement velocity is above the critical gravity drainage velocity, large fractions of water remain in the packed bed. Reflecting the nature of instability, the initial gas saturation varies widely from drainage to drainage. Heterogeneity in porosity enhances the effect of fingering and reduces the displacement efficiency. With an increase in Bond number and a decrease in capillary number, the initial gas saturation of 98% can be achieved at a capillary number of  $3.72 \times 10^{-7}$ .

Next, water was injected in the packed beds to evaluate the residual gas saturation. The residual gas saturation has a peak against the initial gas saturation at approximately 50%.

During imbibition processes, migration of gas is assisted by buoyancy because water is injected vertically upward. For higher initial gas saturations, gas in a continuous phase is hardly disconnected from the continuum.

Finally, gas was injected vertically upward into the packed bed to study the effect of the direction of gas injection with respect to gravity on gas saturation. In the case of the upward injection of gas, gravitational forces and capillary pressure let water remain in porous media against gas migration. With a decrease in the gas injection flow rate, the initial gas saturation increases; however, the initial gas saturation due to upward gas injection is lower than that due to downward gas injection at the same capillary number because of instability of the displacing front.

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