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Epitaxial SrRuO₃ Thin Films Deposited on SrO buffered-Si(001) Substrates for Ferroelectric Pb(Zr_{0.2}Ti_{0.8})O₃ Thin Films

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1. Introduction

At present there is considerable interest in utilizing ferroelectric thin films as a medium for non-volatile data storage [1]. In particular, much attention has been focused on investigating high density giga-bit data storage using scanning probe techniques [2]. It is known that the thickness of a 180° domain wall in ferroelectric thin film is 1-2 nm [3]. As such, these films have potential to serve as a medium for scanning probe microscopy (SPM)-based ultrahigh density (100 Gbit/cm² class) data storage [2]. It has been proposed that employing ferroelectric films as recording media inherently has several advantages: 1) the recorded data are non-volatile, 2) the recording density can be ultra high because of narrow domain wall thickness, 3) the ferroelectric domains can have fast switching speeds, and 4) information bits can be written and read electrically.

Among the various perovskite oxide materials, PZT films appear to be suitable as a storage medium, since their remanent polarizations are high. The highly tetragonal PZT films exhibit high remanent polarization showing a square-shaped hysteresis curve, compared with films having morphotropic and rhombohedral compositions. Therefore, Pb($Zr_{0.2}Ti_{0.8}$)O₃ films with a highly tetragonal structure are suitable for nano-data storage system applications. Ferroelectric Pb($Zr_{0.2}Ti_{0.8}$)O₃ thin films grown epitaxially on SrRuO₃/SrTiO₃ are reported to exhibit large P_r (remanent polarization) [4-5].

SrRuO₃ is a conductive oxide (a room temperature resistivity of 280 μΩ·cm for single crystals), pseudo-cubic perovskite with a lattice constant of 0.55 nm, and has a low mismatch with Pb(Zr_{0.2}Ti_{0.8})O₃(~ 0.5 %) and SrO (~ 1.7 %), thus allowing high quality epitaxial growth. Single-crystal epitaxial thin films of SrRuO₃ and Sr_{1-x}Ca_xRuO₃[6] are frequently grown on SrTiO₃ substrates. In order to achieve epitaxial growth of SrRuO₃ on Si, yttria-stabilized zirconia (YSZ) [7-8] and MgO [9] have been employed as buffer layer materials. Another promising candidate for the buffer layer is SrO which has a NaCl-type cubic structure with a lattice parameter of a = 5.140 Å [10-11]. The SrO(110) dielectric layer also exhibits good compatibility with Si(001) showing a low lattice mismatch of about 0.4 % [12].

In this study, highly tetragonal $Pb(Zr_{0.2}Ti_{0.8})O_3$ films were grown on epitaxial SrRuO₃ thin film electrodes using a SrO buffer layer on Si(001) substrates by pulsed laser deposition. The effect of the SrO buffer layer on the surface morphologies and the electrical properties of the

 $SrRuO_3$ thin film electrodes was investigated as a function of SrO buffer layer thickness. Ferroelectric properties in c-axis oriented $Pb(Zr_{0.2}Ti_{0.8})O_3$ thin films deposited on an epitaxial $SrRuO_3$ electrode were also investigated.

2. Experimental procedure

The SrO target for pulsed laser deposition is difficult to make due to the low sinterability of SrO materials. A SrO₂ target was used to deposit SrO on Si (001) substrates in a vacuum ambient. Before deposition of SrO layers on Si (001) substrates, Si wafers were etched using HF solution to remove the native oxide layers. SrO, SrRuO₃ and PZT films were deposited on etched-silicon substrates using a KrF excimer laser (λ = 248 nm) with a maximum repetition rate of 10 Hz. Laser pulse energy density used for SrO, SrRuO₃ and PZT films deposited using ceramic targets was approximately 1.5 J/cm². 30 mol % excess PbO in the PZT targets was added to compensate for the PbO loss during both the sintering and deposition processes. The distance between the target and substrate was varied depending on the deposited materials. The typical SrO, SrRuO₃, and PZT deposition conditions are summarized in Table 1. The SrO, SrRuO₃, and PZT were in-situ deposited at each temperature in order to ensure chemical stability of the SrO films with Si. The PZT films were cooled down to room temperature at an oxygen pressure of 300 Torr to preserve the oxygen content in the films during the cooling procedure.

Deposition parameters	SrO	SrRuO ₃	$Pb(Zr_{0.2}Ti_{0.8})O_3$
Target	SrO ₂	SrRuO ₃	$Pb(Zr_{0.2}Ti_{0.8})O_3$
Deposition temperature	700°C	550~750°C	575~600°C
Film thickness	3~30nm	50~300nm	100nm
Deposition pressure	1x10-6 Torr	1x10 ⁻² Torr	1x10 ⁻¹ Torr
Energy density	1.5 J/cm ²	1.5 J/cm ²	1.5 J/cm ²
Repetition rate	1 Hz	10 Hz	10 Hz
Target-substrate distance	3 cm	4 cm	6 cm
Substrates	Si (001)	SrO/Si	SrRuO ₃ /SrO/Si

Table 1. Deposition conditions of $Pb(Zr_{0.2}Ti_{0.8})O_3$ thin films, $SrRuO_3$ electrodes, and SrO buffer layers on Si (001) substrates by pulsed laser deposition

The surface morphologies of SrRuO₃ and PZT were characterized by atomic force microscopy (AFM, AUTOPROBE CP, PSI). Crystalline properties of the films were investigated by θ -2 θ , ω -scan, and Φ -scan using a high resolution X-ray diffraction (HRXRD, Rigaku RINT2000). The composition of the SrRuO₃ films was identified by Rutherford backscattering spectroscopy (RBS) using a beam energy of 2.236 MeV (4He⁺⁺) and an incident angle of 160°. The elemental distribution in the PZT/SrRuO₃/SrO/Si structure was investigated by secondary ion mass spectroscopy (SIMS) and Auger electron spectroscopy (AES). The resistivity of SrRuO₃ thin films was measured by an electrometer (CMT-SR 1000) using a four-point probe. The ferroelectric properties and leakage current characteristics were measured using a RT66A ferroelectric tester (Radiant Technology) operating in the virtual ground mode and a Keithley 617 electrometer, respectively. The measurements were carried out using a metal-insulator-metal (MIM) configuration. A Pt top electrode (Area = 7.85 × 10⁻⁵ cm²) patterned using lift-off lithography was prepared at room temperature by dc sputtering.

3. Results and discussion

Figure 1(A) shows XRD patterns of 100 nm thick-SrRuO₃ thin films deposited on SrO buffer layers with various thicknesses. The XRD patterns were plotted using a log-scale to check the existence of the minor portion of SrO₂ phase within the SrRuO₃ films. The SrRuO₃ films were deposited at about 650°C in 1×10^{-2} Torr. As shown in Fig. 1(A), SrRuO₃ films deposited on 6 nm-thick-SrO buffered Si (001) substrates exhibit a c-axis preferred orientation, indicating the (001) and (002) planes alone. In the case of buffer layers above 12 nm thickness, SrO₂ phase was observed in the SrRuO₃ films. From the XRD results, the SrO buffer layer was found to play an important role for the preferred growth of SrRuO₃ films on Si substrates at below approximately 6 nm-thickness. In order to investigate the epitaxial relationship between SrRuO₃ and SrO(6 nm)/Si(001), a Φ -scan in the SrRuO₃ {111} can be observed at every 90°, indicating that the SrRuO₃ films are epitaxially grown on a SrO/Si (001).

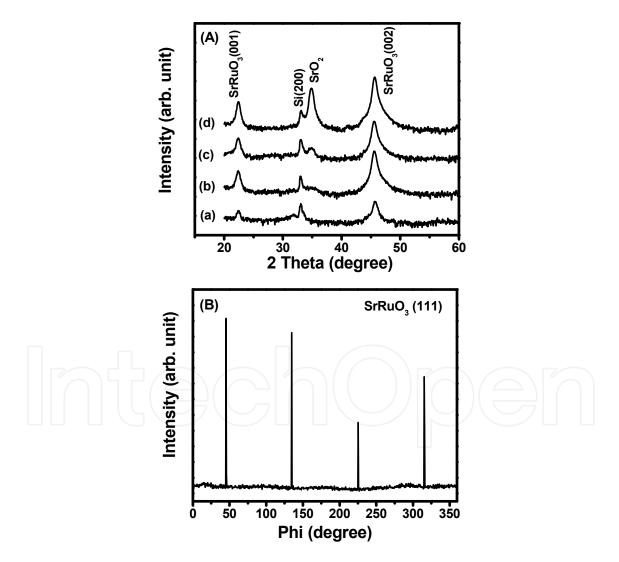


Fig. 1. (A) XRD patterns of 100nm thick-SrRuO₃ thin films deposited on SrO buffer layers of (a) 3 nm, (b) 6 nm, (c) 12 nm, and (d) 30 nm thickness. (B) Φ -scan profile of SrRuO₃ films deposited on SrO (6nm)/Si. (XRD patterns were plotted using a log-scale)

Figure 2 shows the variation in resistivity and rms roughness of 100nm thick-SrRuO₃ thin films as a function of SrO thickness. The SrRuO3 films deposited on ultra-thin SrO films of about 3nm show the highest rms roughness and resistivity values because the SrO films do not play a role as a buffer layer. On the other hand, above 6nm thickness, SrRuO₃ films exhibit a low rms roughness of about 3-5 Å and a resistivity of 1700 – 1900 $\mu\Omega$ -cm. The resistivity values of the SrRuO₃ films deposited on 6nm-thick SrO buffer layers are approximately 1700 $\mu\Omega$ -cm, higher than that of SrRuO₃ films (~ 400 μ Ω-cm) deposited on SrTiO₃ single crystals[13] and (100) LaAlO₃ single crystals[14]. The high resistivity of the SrRuO₃ films on Si substrates originates from the existence of silicon oxide formed from the diffusion of silicon, because the thin SrO layer does not prevent the diffusion of silicon during the deposition of SrRuO₃ at high temperature. The full-width-half-maximum (FWHM) values of the SrRuO₃ films deposited on SrO/Si and SrTiO₃ single crystal substrates are approximately 7.32° and 0.15°, measured by ωscan, respectively. These results suggest that the SrRuO₃ films grown on Si substrates are inferior in terms of film qualities such as crystallinity and defects relative to those grown on SrTiO₃ single crystals. The inset in Fig. 2 shows three-dimensional AFM images of SrRuO₃ films deposited on (a) 3 nm thick- and (b) 30 nm thick-SrO buffer layers.

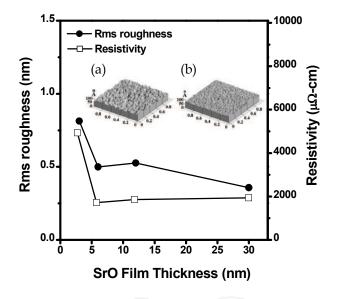


Fig. 2. Variation in resistivity and rms roughness of 100 nm thick-SrRuO₃ thin films as a function of SrO thickness. The inset shows three-dimensional AFM images of SrRuO₃ thin films deposited on SrO buffer layers of (a) 3 nm and (b) 30 nm thickness

Figures 3(a) and 3(b) show XRD patterns and rms roughness and resistivity of SrRuO₃ films, respectively, as a function of SrRuO₃ thickness. The SrRuO₃ films were deposited at 650°C on 6 nm thick-SrO buffer layers. As shown in Fig. 3(a), the peak intensity of SrRuO₃ {001} increases with increasing SrRuO₃ thickness, and the SrRuO₃ films of 50 nm thickness also exhibit a *c*-axis preferred relationship with Si(001) substrates. The resistivity of the SrRuO₃ films exhibits a constant value of about 1700 μ Ω-cm irrespective of the film thickness above 50nm. The rms roughness of the SrRuO₃ films exhibits a similar tendency with the variation of resistivity as a function of SrRuO₃ thickness. The film roughness in conducting materials is inversely proportional to the mobility of the charge carriers. The resistivity in the conducting films is also inversely proportional to the mobility of the charge carriers if the concentration of the charge carriers is constant.

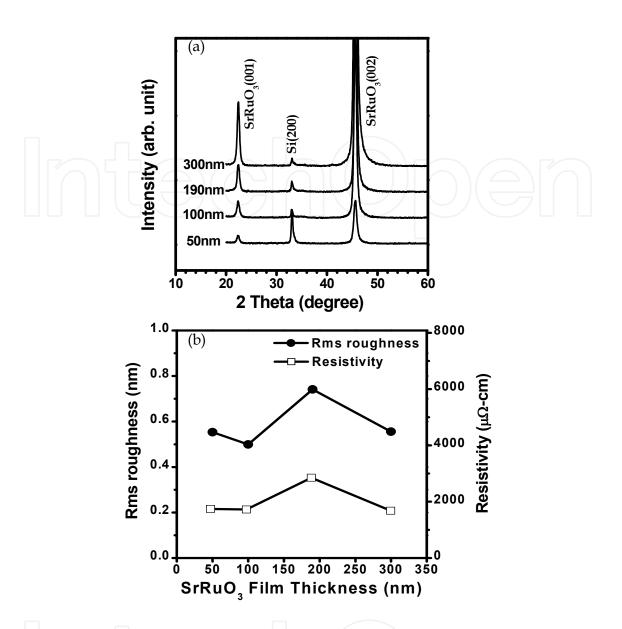


Fig. 3. (a) XRD patterns, (b) rms roughness and resistivity of the SrRuO₃ films deposited on 6nm-thick SrO buffer layers as a function of SrRuO₃ thickness

Figure 4(a) and 4(b) show XRD patterns and variation of rms roughness and resistivity, respectively, for 190nm-thick SrRuO₃ films as a function of SrRuO₃ deposition temperature. As shown in Fig. 4(a), SrRuO₃ films deposited at 550 and 600°C show SrRuO₃ (110) peaks in addition to the SrRuO₃{001} peaks. This indicates that the films deposited at lower temperatures exhibit a polycrystalline nature rather than an epitaxial relationship. The SrRuO₃ (110) peak disappears in the films deposited above 650°C and the films are grown with an epitaxial relationship with Si (001) substrates. As shown in Fig. 4(b), the rms roughness of the films continuously increases with increasing deposition temperature, and a rms roughness of 7.4 Å is noted in the films deposited at 650°C. The rms roughness of the SrRuO₃ films deposited on SrO/Si substrates is higher than that of the films deposited on SrTiO₃ single crystals [13]. In the SrRuO₃/SrO/Si structure, even though the films deposited above 650°C were epitaxially grown, the high rms roughness of the SrRuO₃ films was

attributed to the unstable interface structure of SrO/Si, compared with the SrTiO₃ single crystal substrate. The resistivity of the SrRuO₃ films abruptly decreases as the deposition temperature increases up to 650°C and maintains a constant value of about 1700 $\mu\Omega$ -cm at deposition temperatures above 650°C. The SrRuO₃ films grown with an epitaxial relationship exhibit lower resistivity than the polycrystalline films.

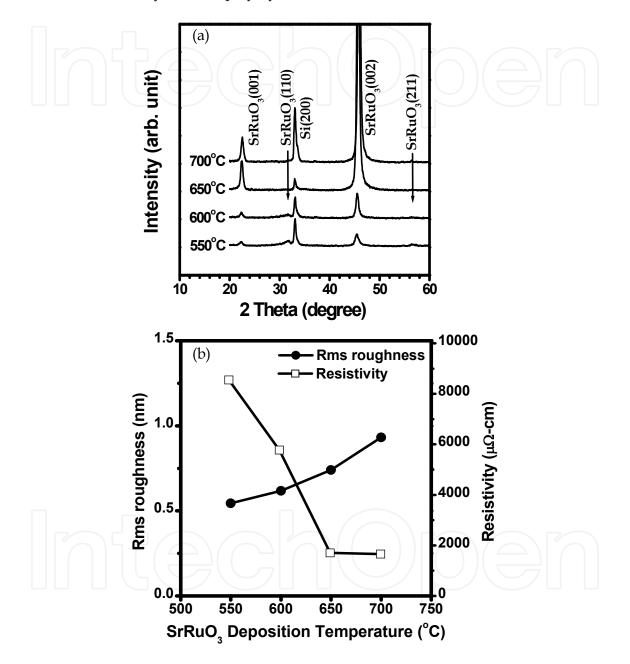


Fig. 4. (a) XRD patterns and (b) rms roughness and resistivity of the SrRuO₃ films deposited on 6nm-thick SrO buffer layers as a function of SrRuO₃ deposition temperature

Figure 5 shows the RBS spectrum of 190 nm thick-SrRuO₃ films deposited at 650°C on SrO/Si substrates. The compositions of Sr and Ru in the SrRuO₃ films deposited from stoichiometric SrRuO₃ targets are Sr/(Sr+Ru) = 50.52 % and Ru/(Sr+Ru) = 49.48 % from the RBS analysis. Because the Sr and Ru elements are overlapped, the composition of the films

was fitted by repeated adjustments with the standard composition of SrRuO₃. The results suggest that the epitaxial SrRuO₃ films deposited by pulsed laser deposition show the same composition as SrRuO₃ targets.

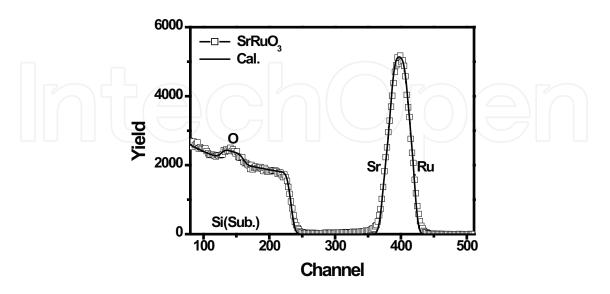


Fig. 5. RBS spectrum of 190nm-thick SrRuO₃ films deposited at 650°C on SrO/Si substrates

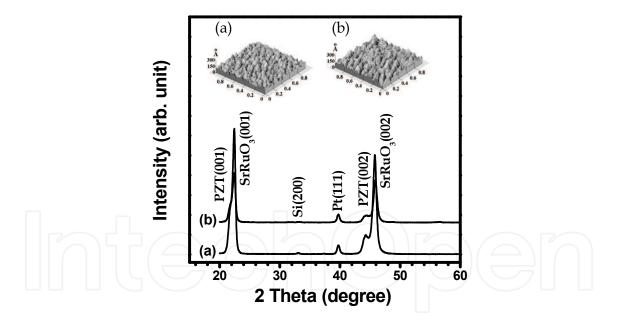


Fig. 6. XRD patterns of PZT thin films deposited at (a) 575° C and (b) 600° C on 190nm-thick SrRuO₃ films. The insets show three-dimensional AFM images of PZT films deposited at (a) 575° C and (b) 600° C

Figure 6 shows XRD patterns of 100 nm thick-Pb($Zr_{0.2}Ti_{0.8}$)O₃ (PZT) films deposited at 575 and 600°C on epitaxial SrRuO₃/SrO/Si substrates. The (a) and (b) in inset of Fig. 6 show three dimensional AFM images of PZT films deposited at 575°C and 600°C, respectively. The rms roughness of the PZT films deposited at 575 and 600°C are approximately 23Å and 26Å, respectively. The PZT films, as revealed in the XRD patterns of Fig. 6, exhibit a close

relationship with SrRuO₃{001} at deposition temperatures of 575 and 600°C. The Pt (111) peak originates from the top electrode in the Pt/PZT/SrRuO₃/SrO/Si capacitor structures. However, the peak intensities of PZT{001} decrease with increasing deposition temperature. The PZT films react with the SrRuO₃ bottom electrode at high deposition temperature, resulting in Pb diffusion into the SrRuO₃/SrTiO₃ [15]. Even though Pb is diffused into the SrRuO₃ films, PZT films deposited on SrRuO₃/SrTiO₃ exhibit a good epitaxial relationship maintaining the predominating crystallinity. In order to investigate the distributions of each element in the PZT films deposited on SrRuO₃/SrO/Si, the elemental distribution in each layer was analyzed by secondary ion mass spectroscopy (SIMS), as shown in Fig. 7 (a). The Pb from the PZT films deposited at 600°C was clearly observed within the SrRuO₃ layer, indicating similar diffusion behavior of Pb into the SrRuO₃/SrTiO₃ [15]. In addition, silicon was also observed at the PZT layer as well as at the SrRuO₃ layer. The silicon existing in the PZT and SrRuO₃ films will present as silicon oxide, because silicon oxide phase is thermodynamically stable compared with Si element. The silicon oxide will exert a harmful influence upon the crystallinity and the morphologies of the PZT films. The existence of silicon in the PZT layer and Pb in the SrRuO₃ layer was also verified by the AES depthprofile as shown in Fig. 7(b). A phi-scan was performed to identify whether the PZT films are epitaxially grown on the SrRuO₃ films. From the results (not shown here), PZT films do not exhibit epitaxial growth on the SrRuO₃ bottom electrode. The PZT films were only grown with (001) preferred orientation on SrRuO₃ electrodes. From the ω-scan of the PZT films deposited at 575°C, the FWHM value of the PZT (002) is approximately 7.08°. Thus, the crystalline quality of the PZT films deposited on SrRuO₃/SrO/Si was distinctly influenced by the silicon oxide within the PZT layers.

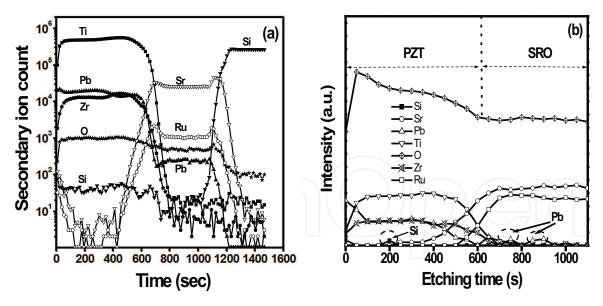


Fig. 7. (a) SIMS and (b) AES depth-profiles of PZT (100nm) films deposited at 600° C on SrRuO₃/SrO/Si

Figures 8(a) and 8(b) show P-E hysteresis loops and leakage current densities, respectively, as a function of applied voltage in 100 nm thick-PZT films deposited at 575 and 600°C. The PZT films deposited at both temperatures show similar P-E hysteresis loops, and were polycrystalline nature. The $2P_r$ (remanent polarization) and E_c (coercive field) of the PZT films are approximately 40 μ C/cm² and 100 kV/cm, respectively. The lower P_r values

relative to reported values may be due to the poor crystallization of the PZT films resulting from inclusion of silicon oxide phase. As shown in Fig. 8(b), leakage current densities of the PZT films deposited at 575°C are approximately 2×10^{-7} A/cm² at 1 V. The breakdown strength of the films was approximately 150kV/cm. The lower breakdown strength of the films may be due to the rough surface morphologies, as shown in the inset of Fig. 6.

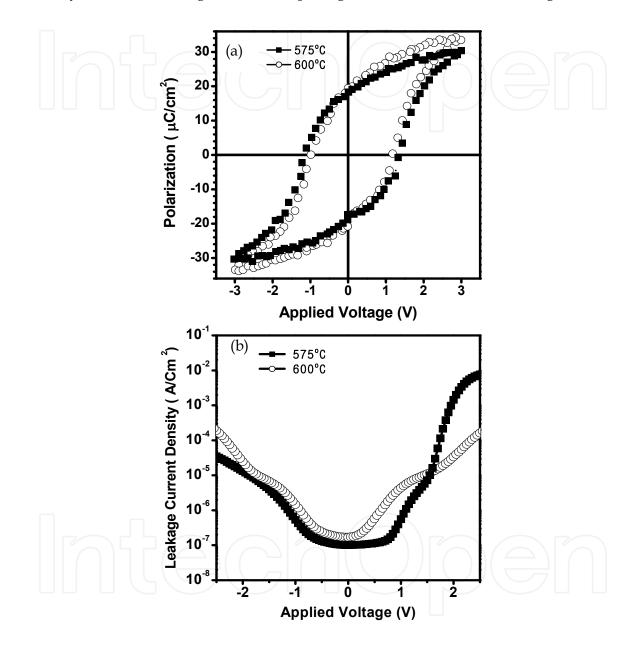


Fig. 8. (a) P-E hysteresis loops and (b) leakage current densities of 100nm thick-PZT thin films deposited at 575° C and 600° C

4. Conclusions

 $SrRuO_3$ bottom electrodes were grown with an epitaxial relationship with SrO buffered-Si(001) substrates by pulsed laser deposition. The structural and electrical properties of the $SrRuO_3$ films were studied with deposition parameters of $SrRuO_3$ on the optimized SrO

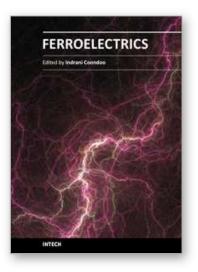
buffer layered Si (001) substrates. The optimum conditions of SrO buffer layers for SrRuO₃ preferred growth were a deposition temperature of 700°C, deposition pressure of 1×10^{-6} Torr, and thickness of 6 nm. 100nm thick-SrRuO₃ bottom electrodes deposited at 650°C on SrO buffered-Si(001) substrates showed a rms roughness of approximately 5.0 Å and a resistivity of 1700 µΩ-cm. 100nm thick-Pb(Zr_{0.2}Ti_{0.8})O₃ thin films deposited at 575°C on SrRuO₃/SrO/Si substrates showed a (00*l*) preferred orientation and exhibited a 2P_r of 40 µC/cm² and a E_c of 100 kV/cm. The leakage current density of the PZT films was approximately 1×10^{-7} A/cm² at 1 V. The silicon oxide phase, which presents within the PZT and SrRuO₃ films, influences the crystallinity of the PZT films and the resistivity of the SrRuO₃ electrodes.

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Ferroelectric materials exhibit a wide spectrum of functional properties, including switchable polarization, piezoelectricity, high non-linear optical activity, pyroelectricity, and non-linear dielectric behaviour. These properties are crucial for application in electronic devices such as sensors, microactuators, infrared detectors, microwave phase filters and, non-volatile memories. This unique combination of properties of ferroelectric materials has attracted researchers and engineers for a long time. This book reviews a wide range of diverse topics related to the phenomenon of ferroelectricity (in the bulk as well as thin film form) and provides a forum for scientists, engineers, and students working in this field. The present book containing 24 chapters is a result of contributions of experts from international scientific community working in different aspects of ferroelectricity related to experimental and theoretical work aimed at the understanding of ferroelectricity and their utilization in devices. It provides an up-to-date insightful coverage to the recent advances in the synthesis, characterization, functional properties and potential device applications in specialized areas.

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